



Department of Chemical
Engineering
University of Texas@Austin



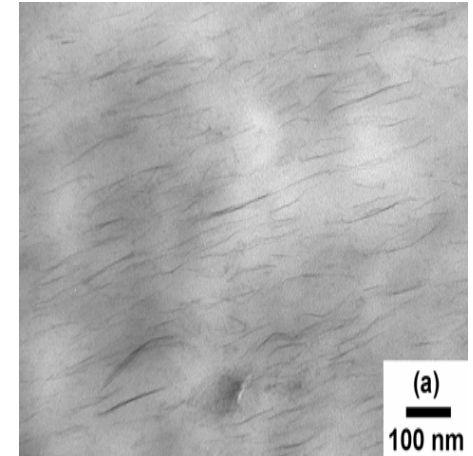
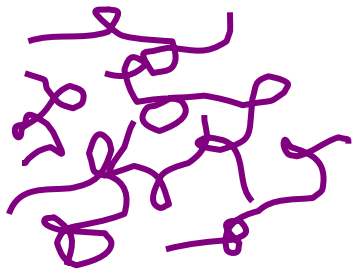
Origins of Mechanical and Rheological Properties of Polymer Nanocomposites

Venkat Ganesan

\$\$\$: NSF DMR, Welch Foundation

Megha Surve, Victor Pryamitsyn

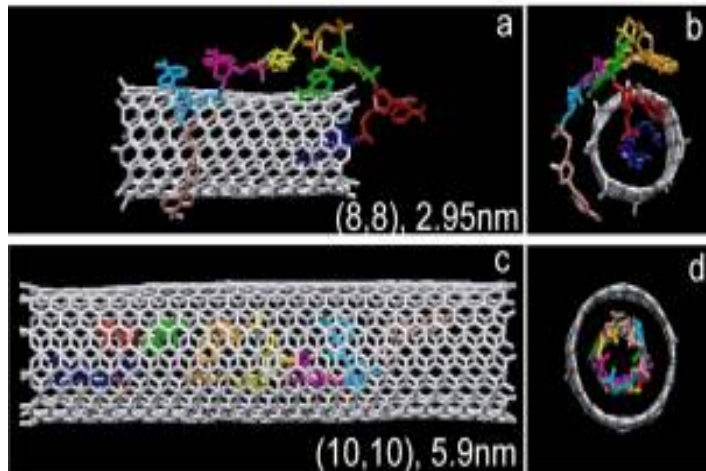
Polymer Nanocomposites



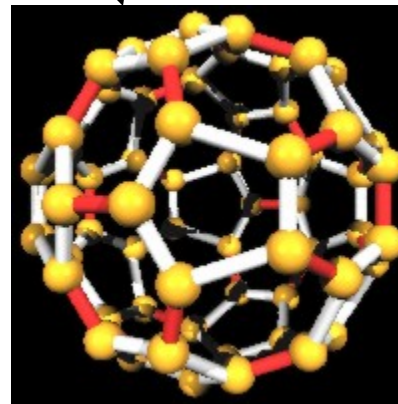
Polymers (Blends,
Block copolymers)

Nano-Fillers

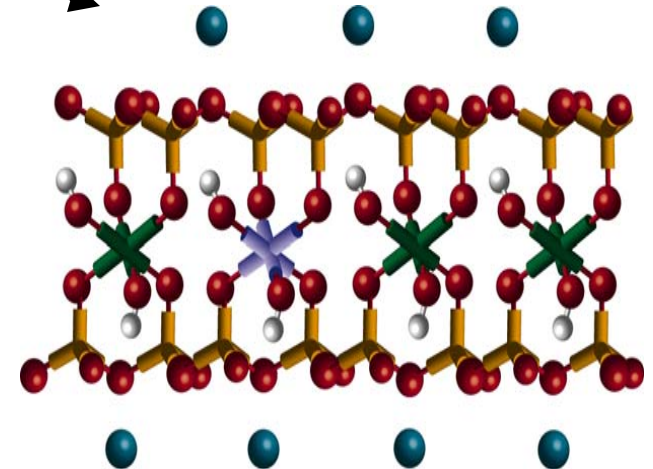
Nanocomposites



Single- and Multi-Walled
Carbon Nanotubes

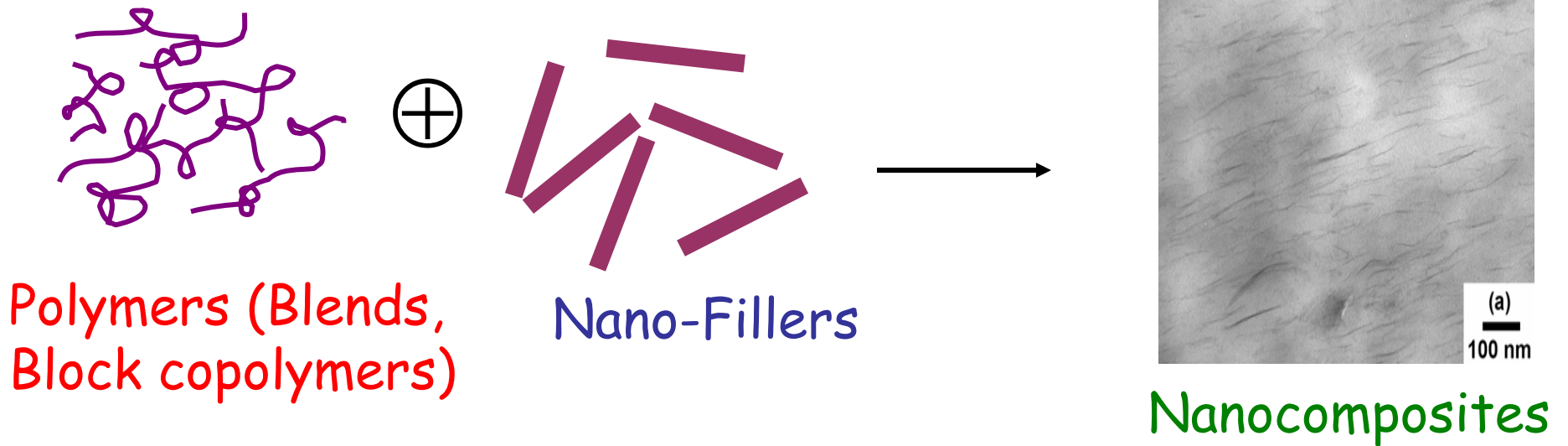


Fullerenes/
Buckyballs



Montmorillonite Clays

Polymer Nanocomposites



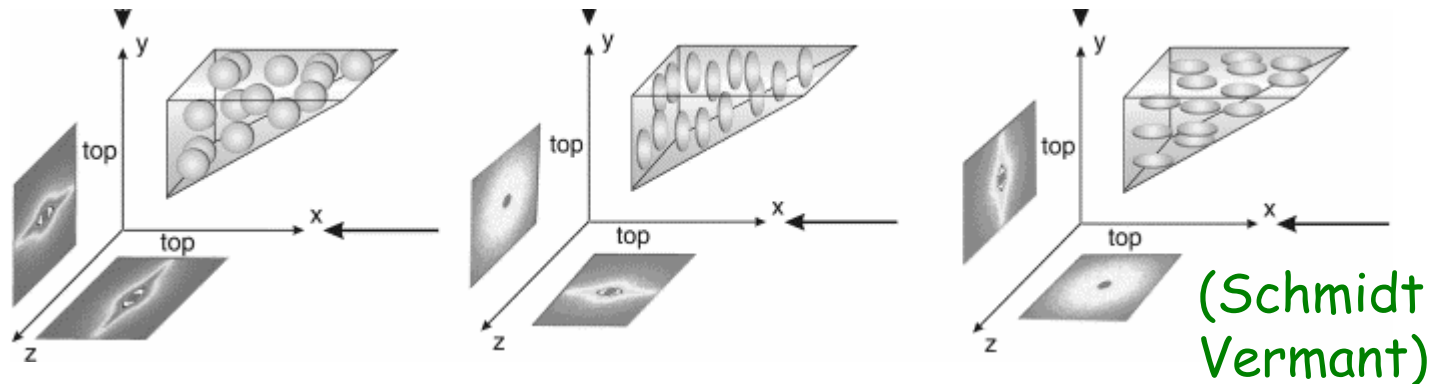
- Processing nanocomposites requires understanding their flow behavior.
- Flow fields provide a versatile approach for controlling dispersions of nanocomposites.

Challenges

- **Objective:** To model the flow dynamics and structure of nanoparticle-polymer mixtures.

Interplay of hydrodynamics and friction in a viscoelastic medium

Flow-induced orientation



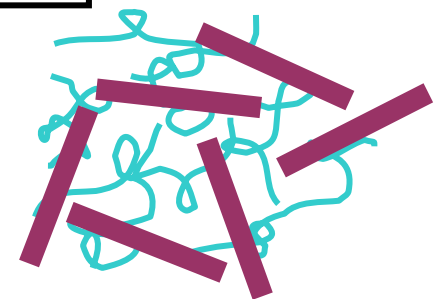
Effect of molecular interactions

Flow-induced gelation



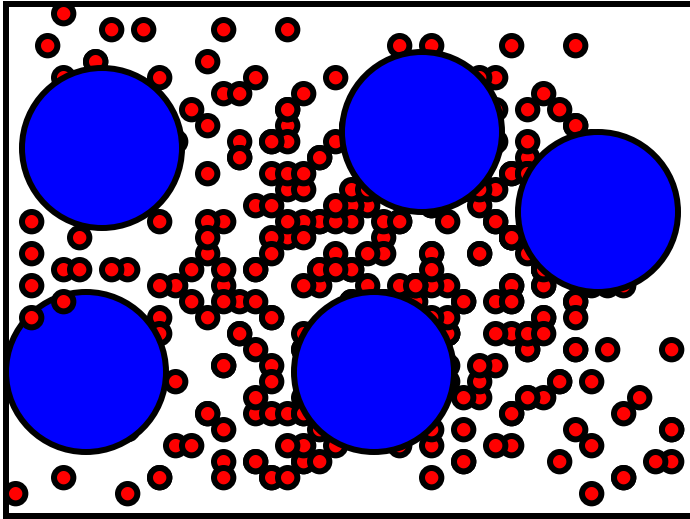
(Schmidt)

Flow-induced Jamming



(Galgali)

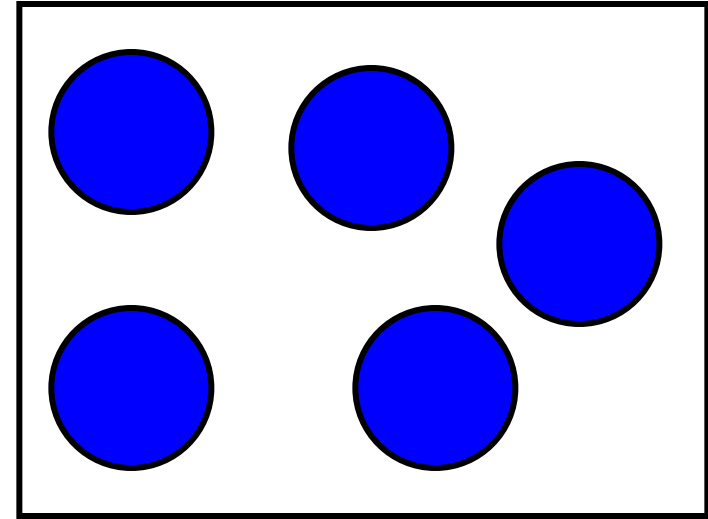
The Approach



Explicit Solvent Method

(Molecular dynamics)

- Captures hydrodynamics and other interactions.
- Due to size asymmetry, is computationally expensive.



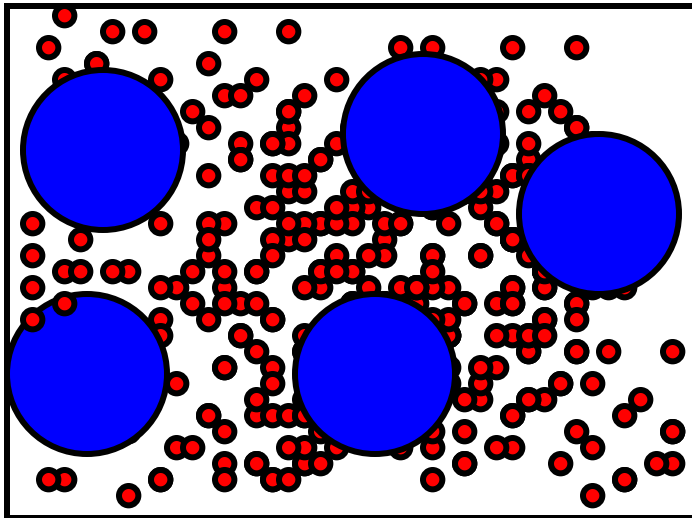
Continuum Methods

(Stokesian dynamics, Lattice-Boltzmann)

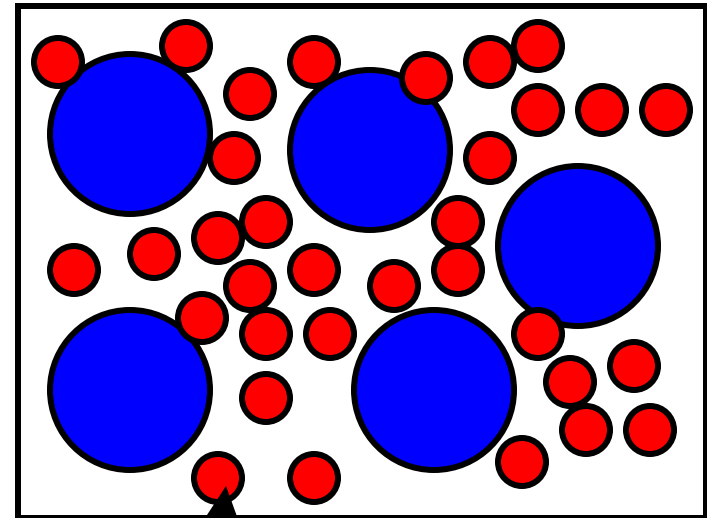
- Captures hydrodynamics and is computationally tractable.
- Can't include interactions with the solvent.
- Not developed for Non-Newtonian flows.

The Approach

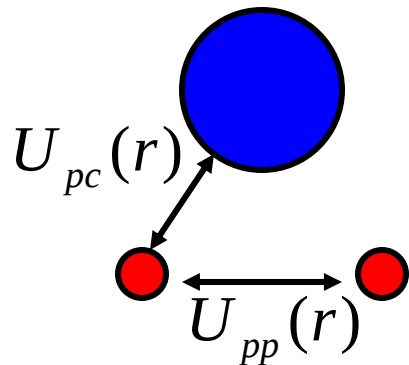
Explicit Solvent Method



Coarse-Grained
Explicit Solvent Method



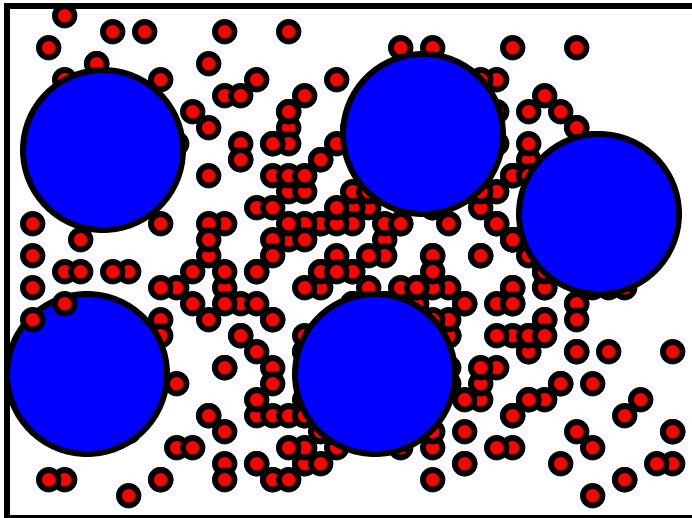
Collection of microscopic solvent units



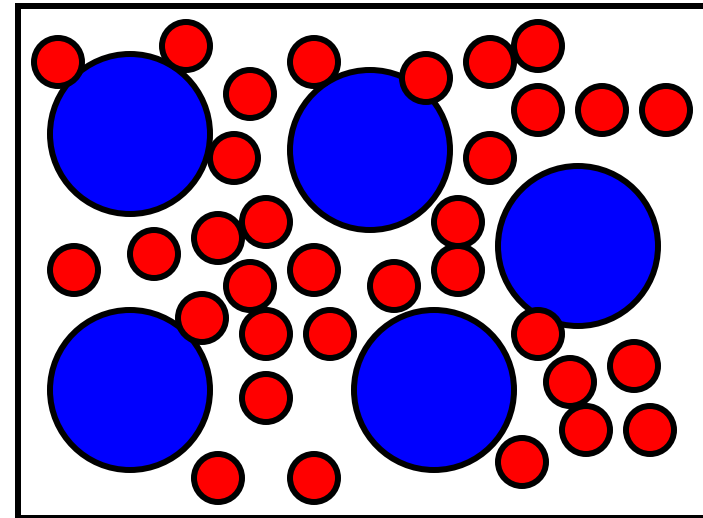
- Particle and solvent units interact by coarse-grained potentials.
- $U_{pc}(r), U_{pp}(r)$ are derivable from more atomistic representations.

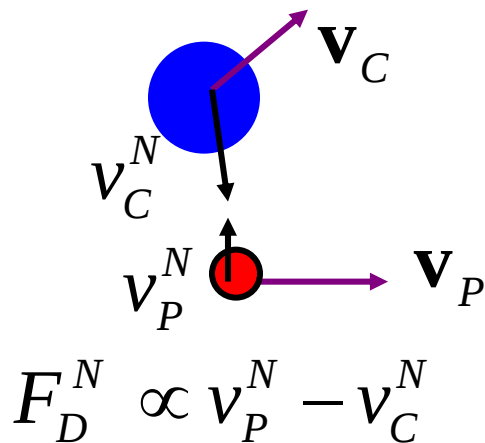
The Approach

Explicit Solvent Method



Coarse-Grained
Explicit Solvent Method

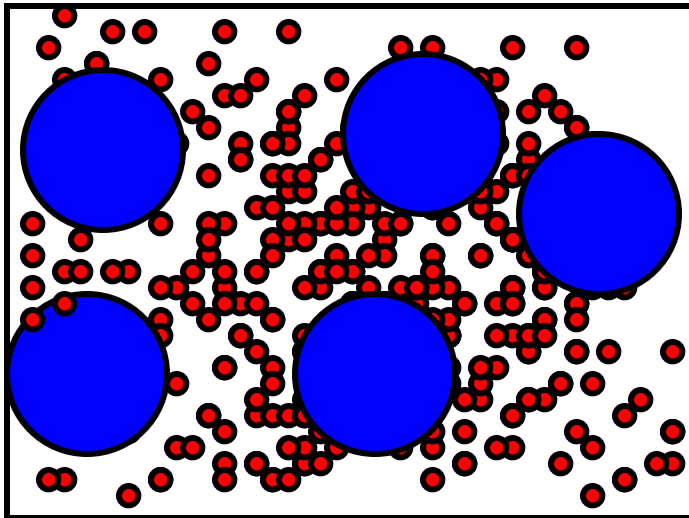



$$F_D^N \propto v_P^N - v_C^N$$

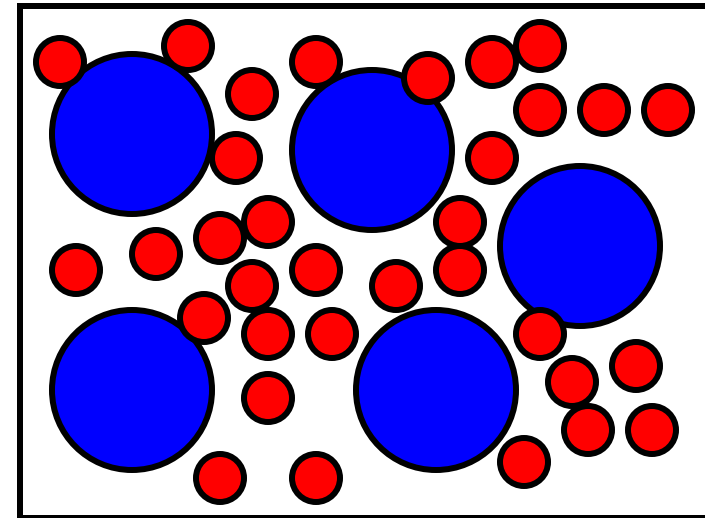
- Particles interact by momentum conserving thermostat (preserves hydrodynamics).
- Involves (central) dissipative forces dependent upon the normal component of the velocity differences.
- Similar to Dissipative Particle Dynamics.

The Approach

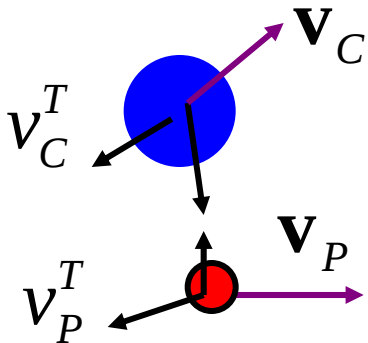
Explicit Solvent Method



Coarse-Grained
Explicit Solvent Method



No-Slip

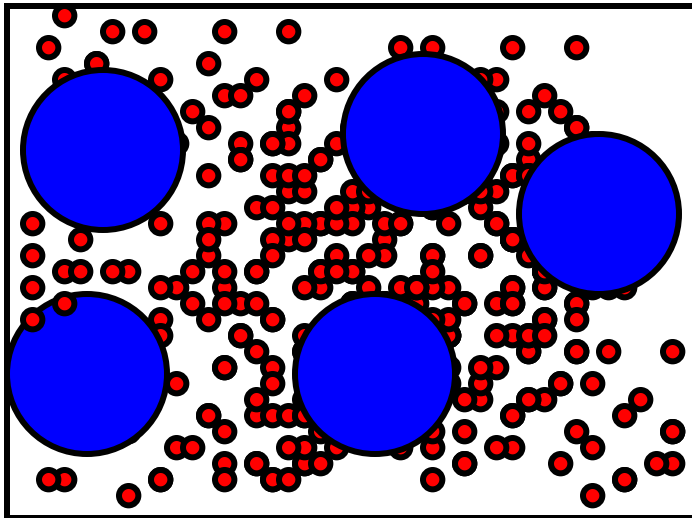


$$F_D^T \propto v_P^T - v_C^T$$

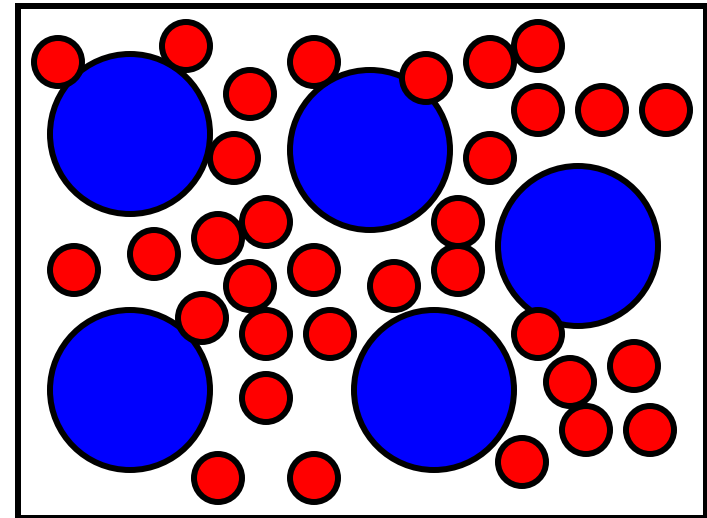
- Does not capture local hydrodynamics.
- Requires tangential (not central) velocity-dependent forces.

The Approach

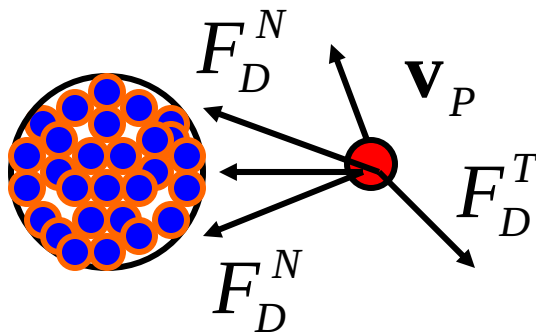
Explicit Solvent Method



Coarse-Grained
Explicit Solvent Method



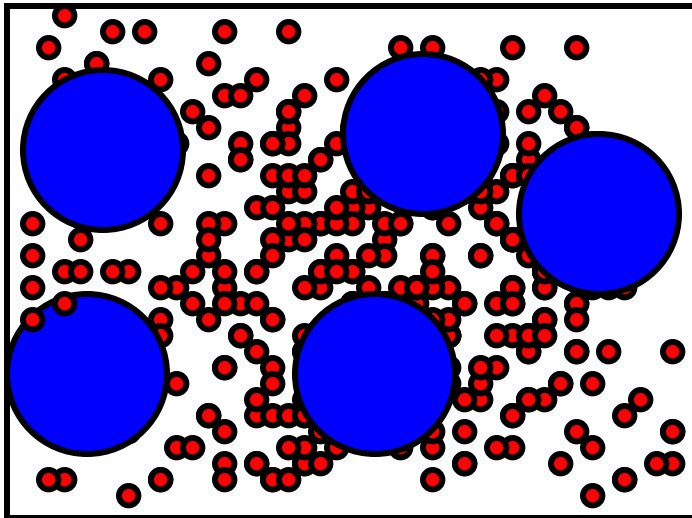
Composite Particles



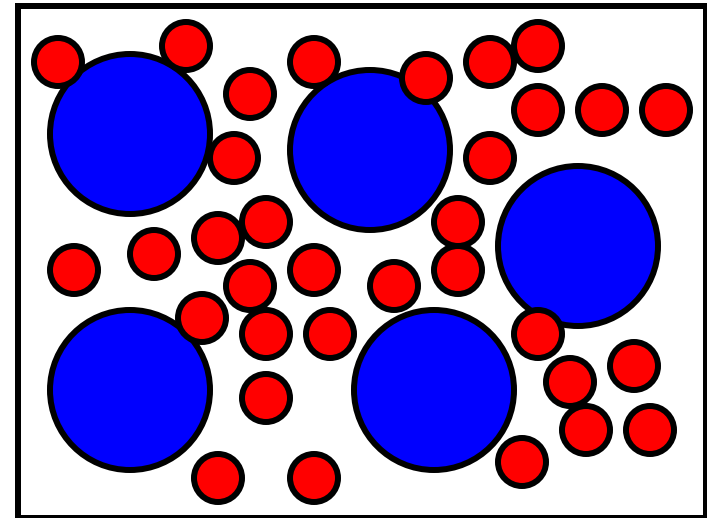
- Does not capture local hydrodynamics.
- Requires tangential (not central) velocity-dependent forces.

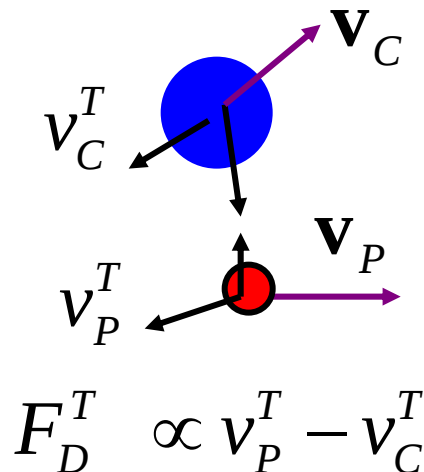
The Approach

Explicit Solvent Method



Coarse-Grained
Explicit Solvent Method

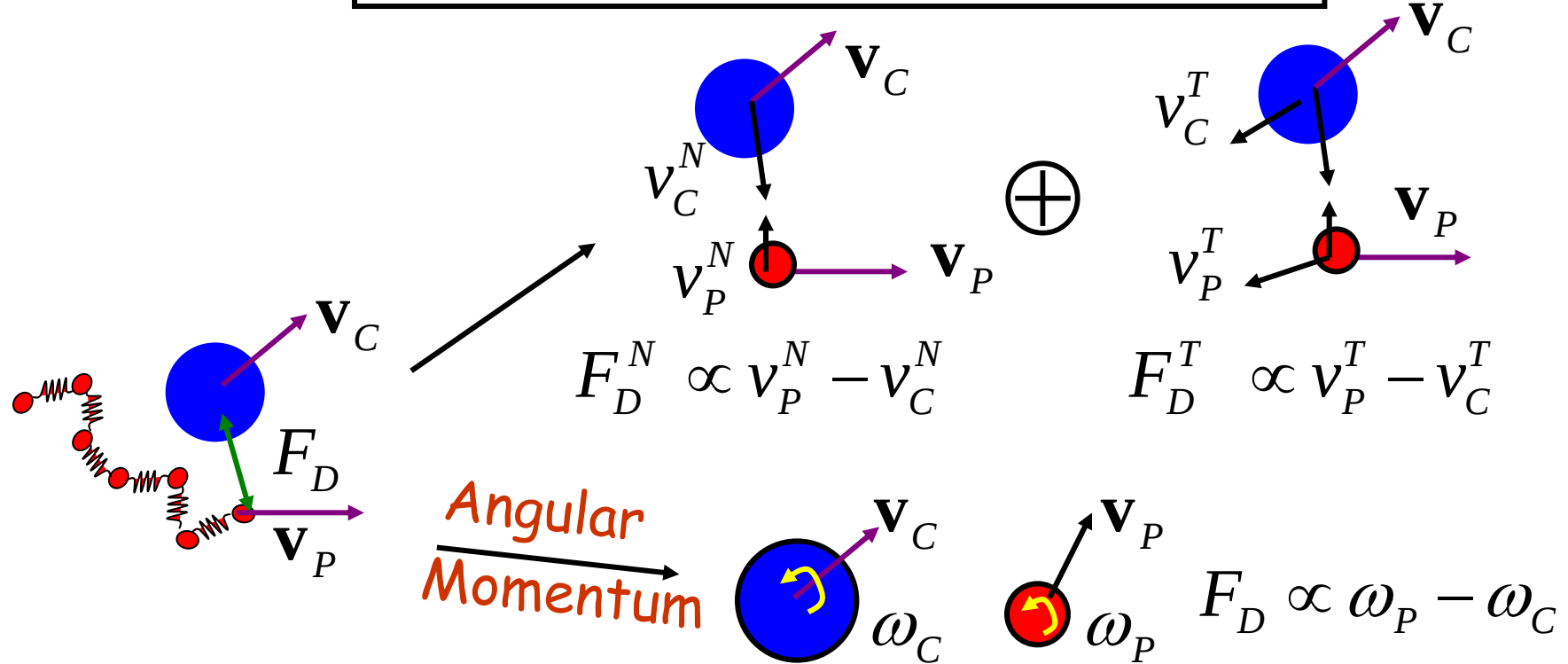



$$F_D^T \propto v_P^T - v_C^T$$

Our proposal

- Directly incorporate tangential velocity-dependent forces.

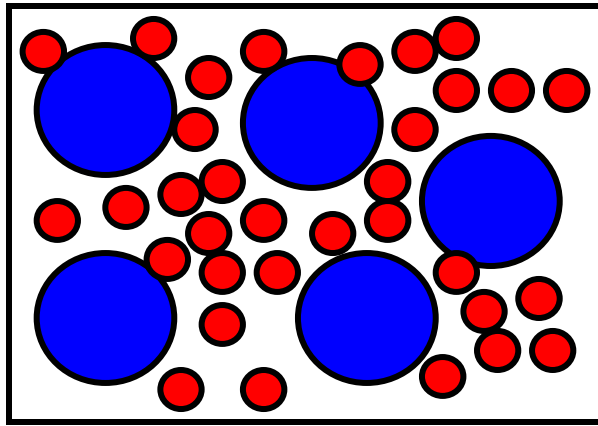
Hydrodynamic Friction Forces



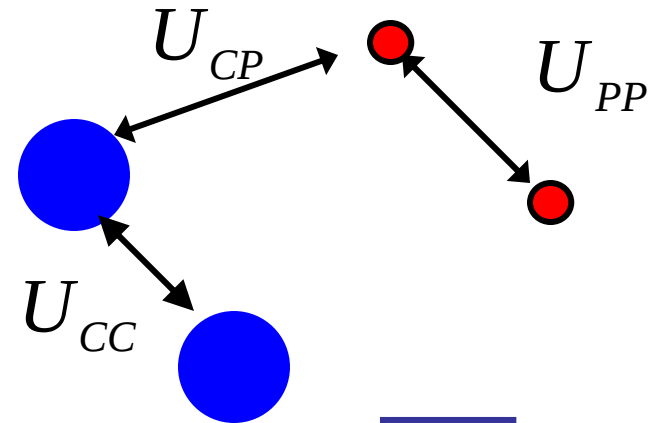
- Conserves linear and angular momentum
 - Preserves hydrodynamical phenomena
 - Includes tangential friction
 - Brownian dynamics + Dissipative forces
 - Computationally tractable (for size asymmetric systems)
- (Espanol, 1998; Pryamitsyn and Ganesan, JCP, 2005)
- U_{ij} Conservative
 F_D^T Dissipative
 F_R Random forces

Coarse-Grained Colloidal Suspension

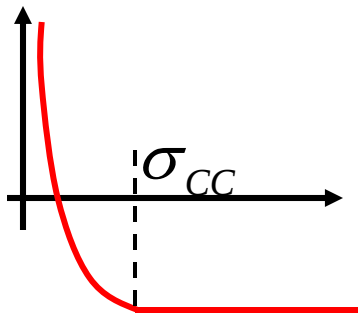
- Mixture of colloid and solvent
- Colloid:Solvent Radius = 5:1.



$\equiv$

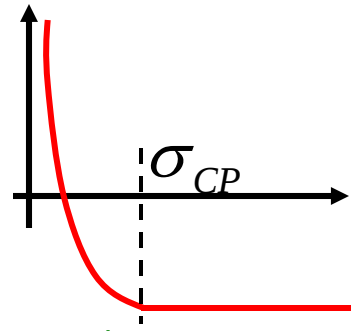


U_{CC}



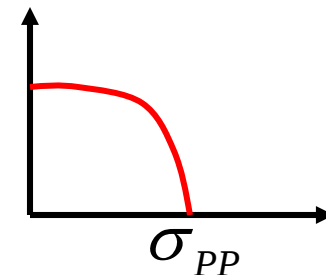
Repulsive LJ

U_{CP}



Repulsive LJ

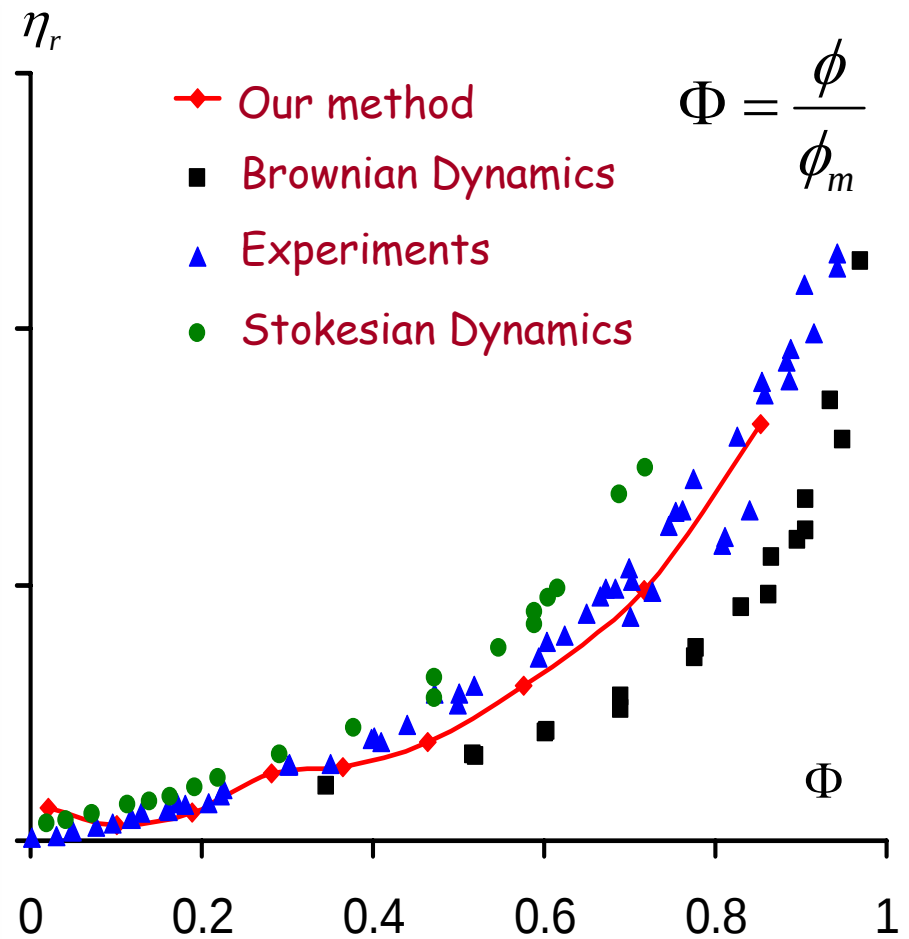
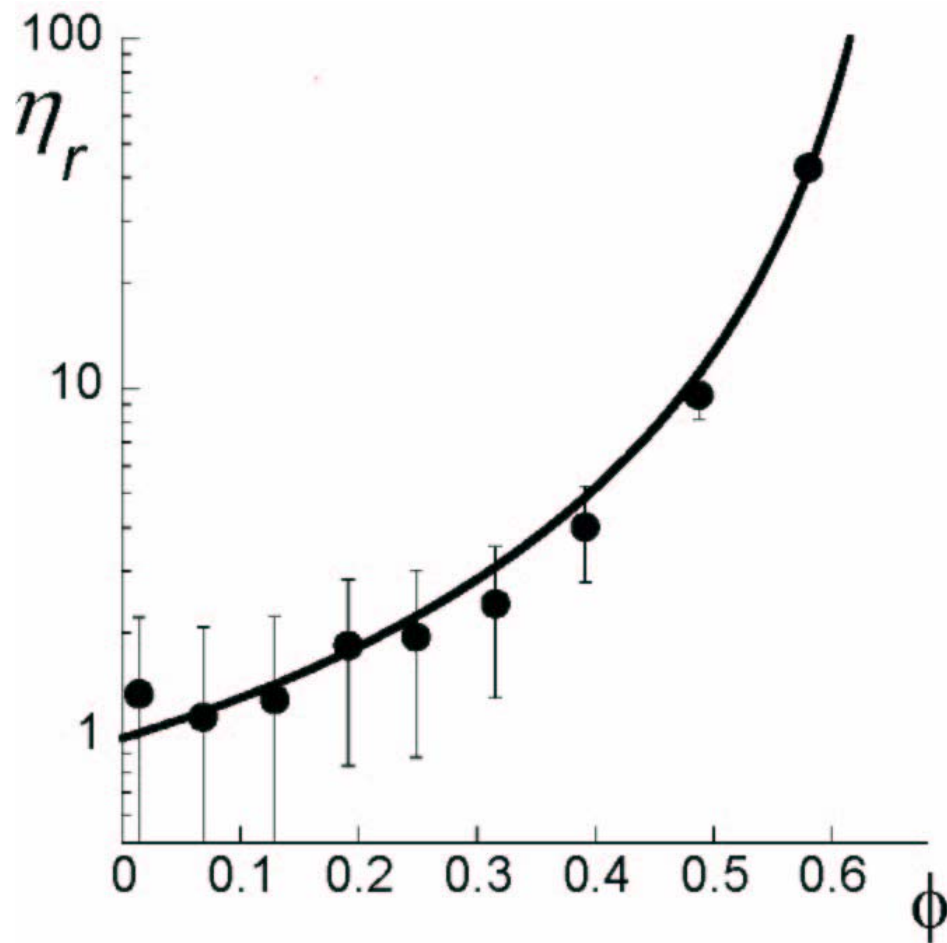
U_{PP}



Soft Repulsion

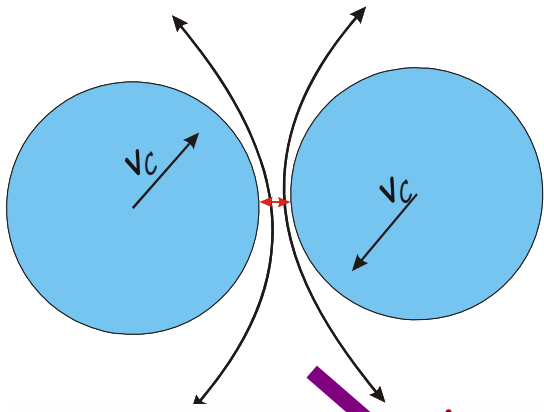
(Pryamitsyn and Ganesan, JCP, 2005)

Hydrodynamic Interactions: Zero-Shear Viscosity

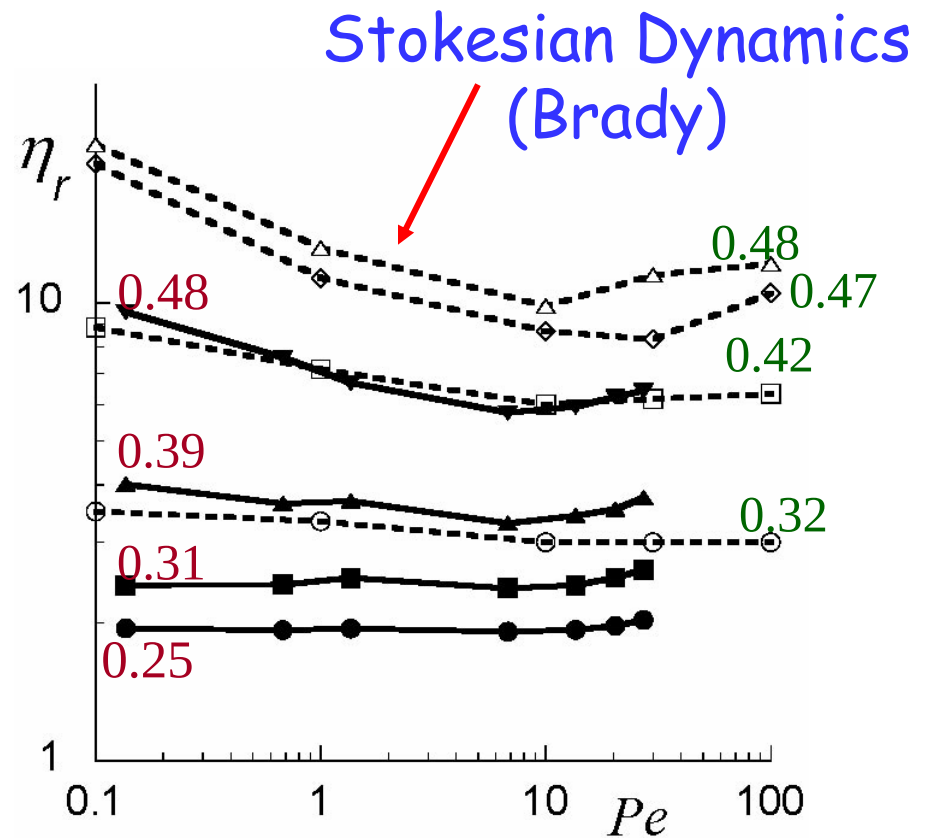
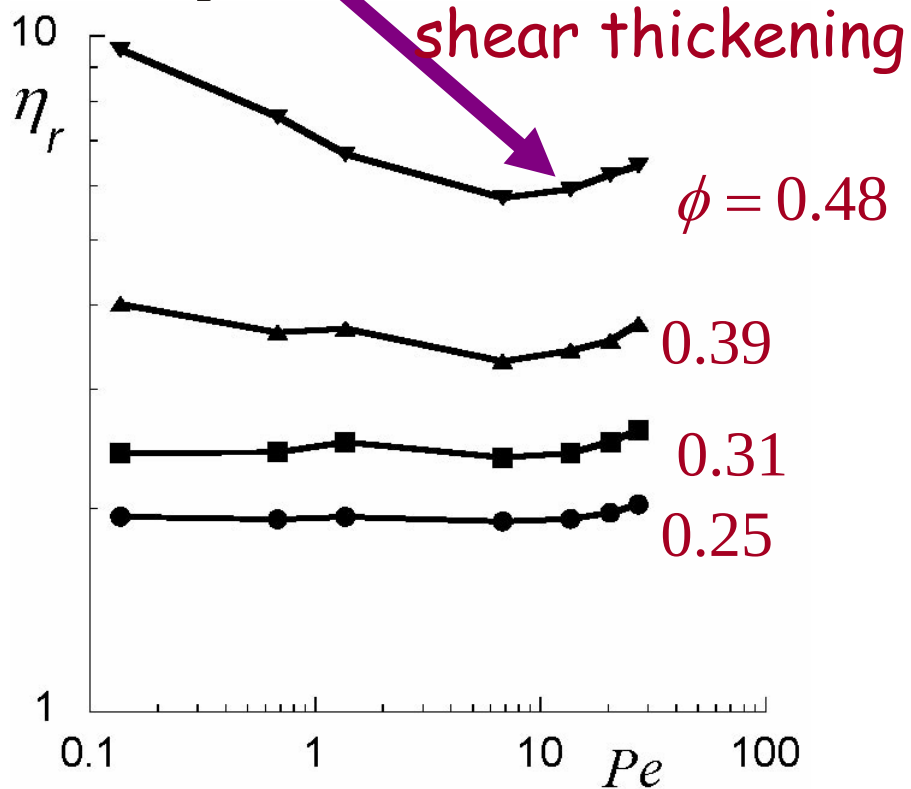


- Coarse-Grained solvent method captures hydrodynamical interactions

Shear Rheology



Rheology is sensitive to lubrication forces
Provides a sensitive test of explicit solvent model.



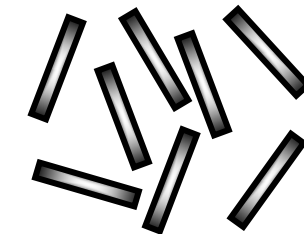
Agrees to within 20% of Stokesian dynamics results

Summary So Far..

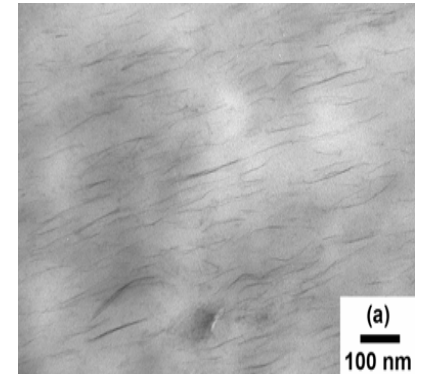
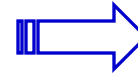
- Outlined a coarse-grained explicit solvent method to simulate hydrodynamical phenomena involving particles in complex fluids.
- Provided evidence that both hydrodynamical and other interactions can be faithfully captured.
- Results on hard sphere suspensions provided new insights into the interplay between glass transition, hydrodynamics and rheology.

Why Polymer Nanocomposites ?

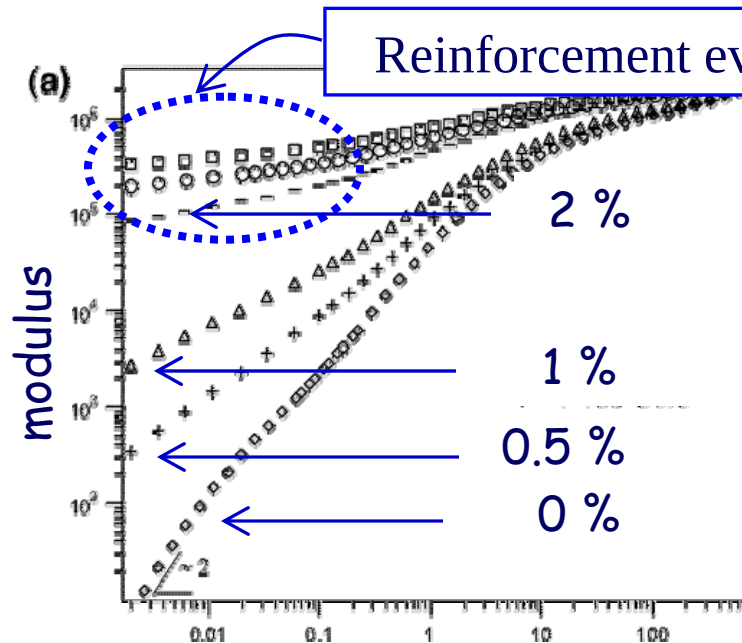
Polymers
(Solutions, Blends)



Nano-fillers (Clay,
Nanotubes, Fullerenes)



Nanocomposites



Silica + PEO[#]

Elastic modulus doubles

Viscosity increases by an order
of magnitude

Microsized particles vol $\sim 30\%$

Nanoparticles vol $\sim 1 - 5 \%$

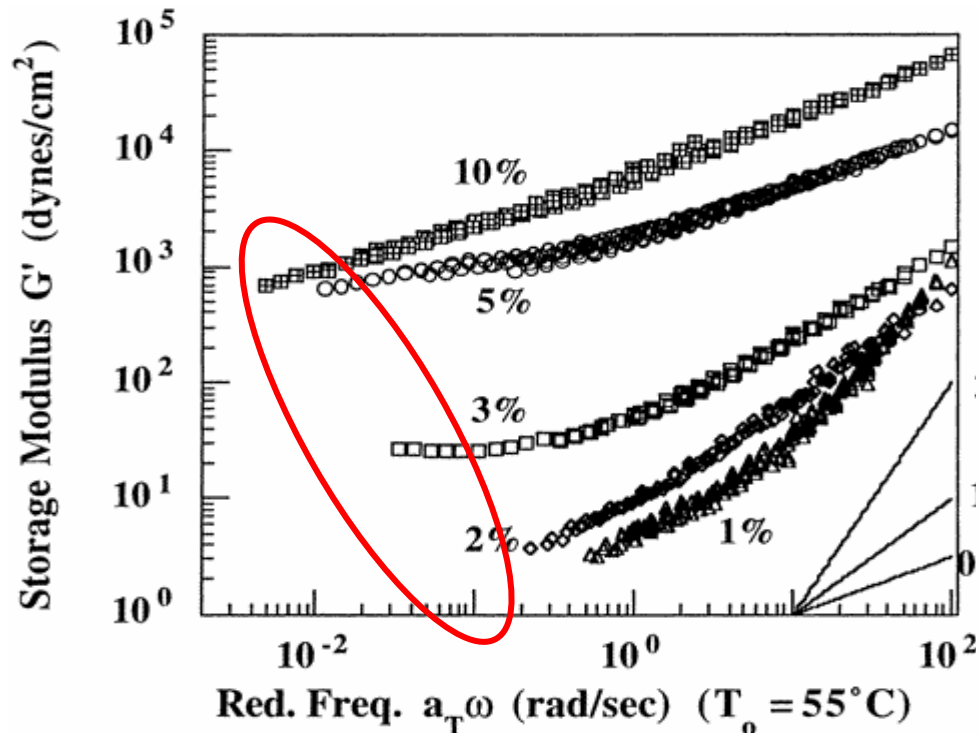
[#] Zhang and Archer, Langmuir, 2002

✓ Addition of small particles - Significant property enhancement!!

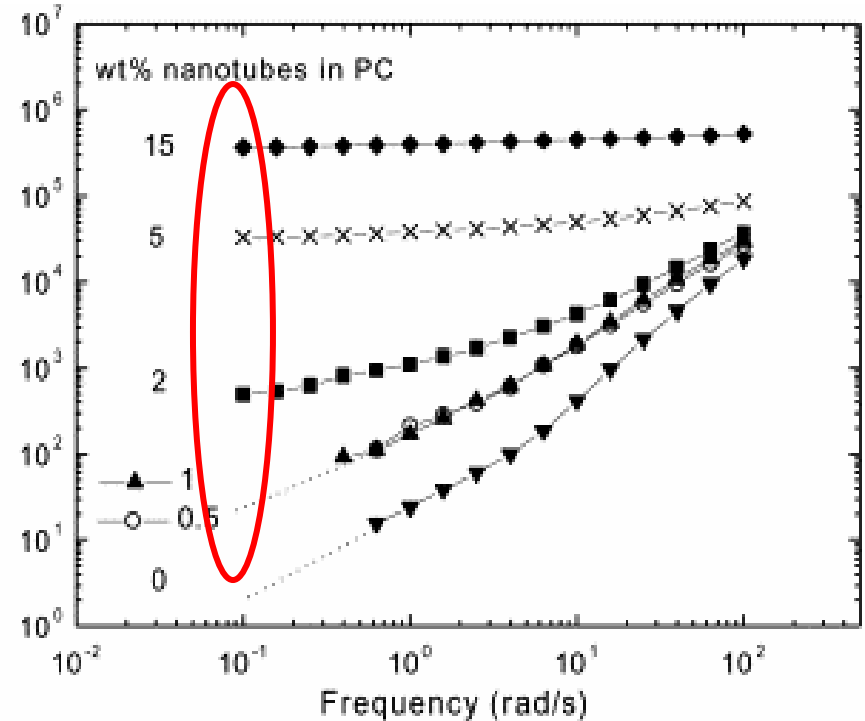
Issues and Questions

- How do nanoparticles modify the mechanical properties of polymer matrices ?
- What are the mechanisms underlying the above effects ?
- What are the parameters governing the mechanisms ?

Rheology of PNCs: Linear Viscoelasticity



Polycaprolactone + Montmorillonite Clays*



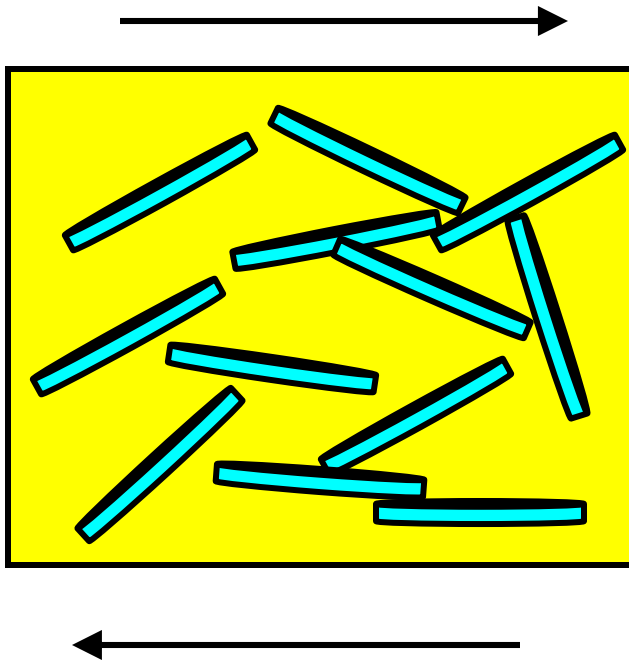
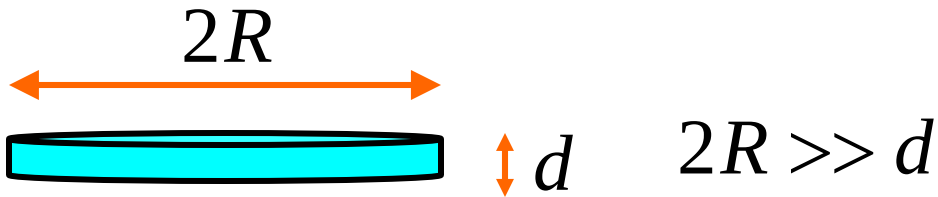
PC + Nanotubes**

- Significant enhancements in elasticity at extremely low loadings
- Change of viscoelastic response to "solid-like" behavior.

*: Krishnamoorti and Giannelis; **: Fornes and Paul

Rheology: Explanations

Particle Jamming/Percolation (Krishnamoorti)



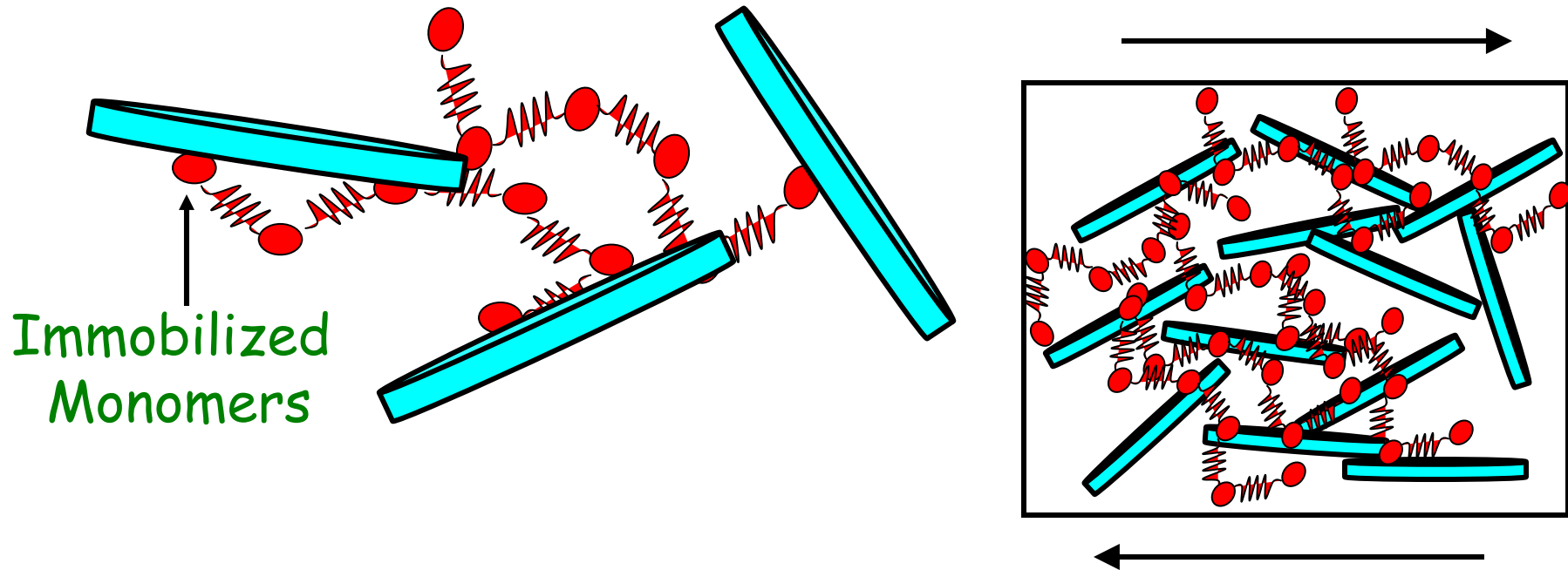
- Jamming/percolation occurs at low ϕ

$$\phi R^3 \approx 1$$

- Leads to solid behavior and the enhancements in modulus.

Rheology: Explanations

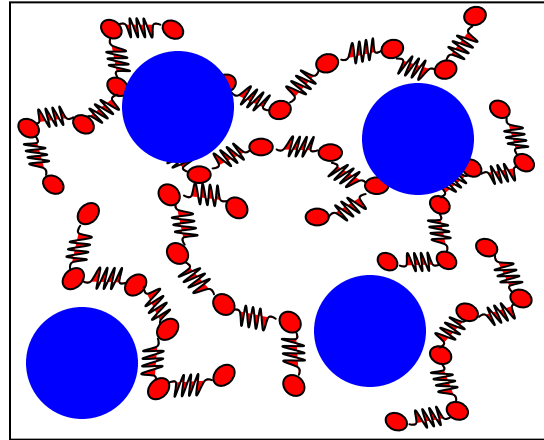
Polymer Network Mechanism (Kumar and Douglas)



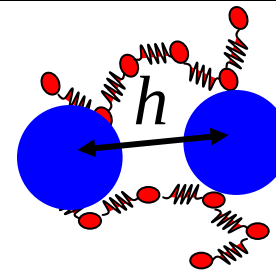
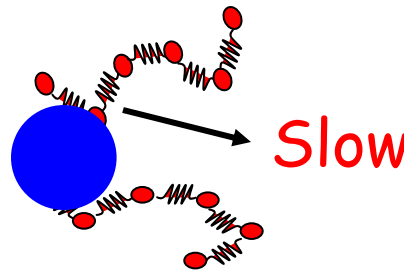
- Elasticity due to transient network formation.
- Plateau modulus due to bridges.

Rheology of PNCs: Model System

- Mixture of spherical nanoparticles in polymer matrices



$$\frac{2R_g^P}{\sigma_{CC}} \approx O(1)$$

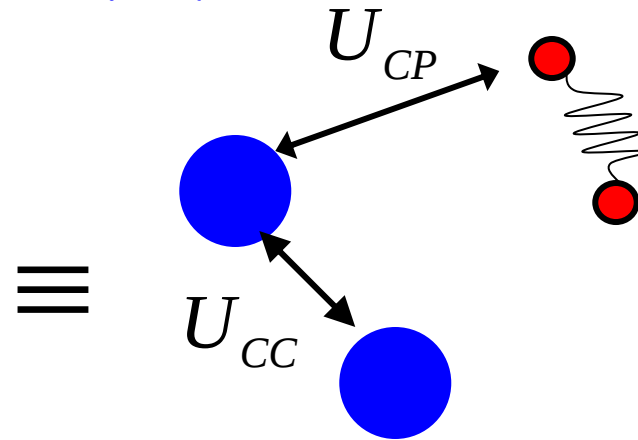
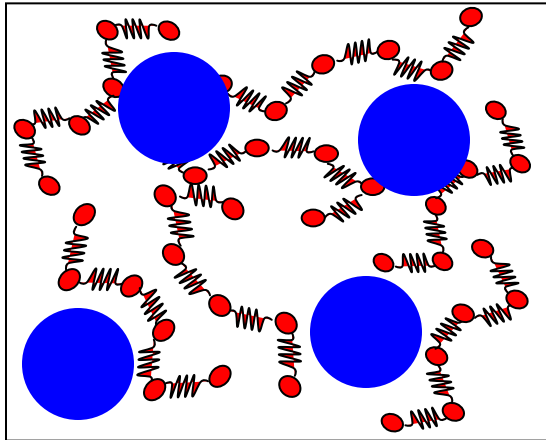


$$\frac{h}{R_g} \approx O(1)$$

- **Advantage:** A lot is known about spherical colloidal dispersions
- **Disadvantages:** Orientational effects are absent.
Need much higher loadings.

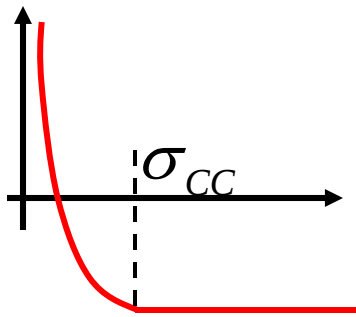
Model of Polymer Nanocomposites

- Mixture of colloid and polymeric melt



N_P	$\frac{2R_g^P}{\sigma_{CC}}$
4	0.45
48	1.6

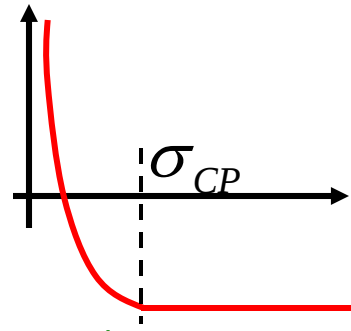
U_{CC}



Repulsive LJ

$$\frac{\sigma_{CC}}{\sigma_{PP}} = 5.0$$

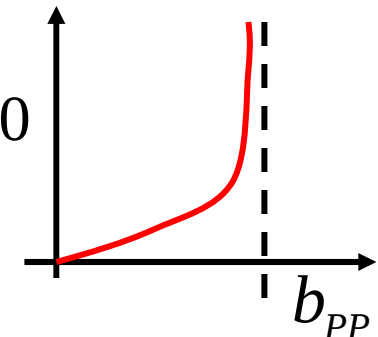
U_{CP}



Repulsive LJ

$$\frac{b_{PP}}{\sigma_{PP}} = 3.0$$

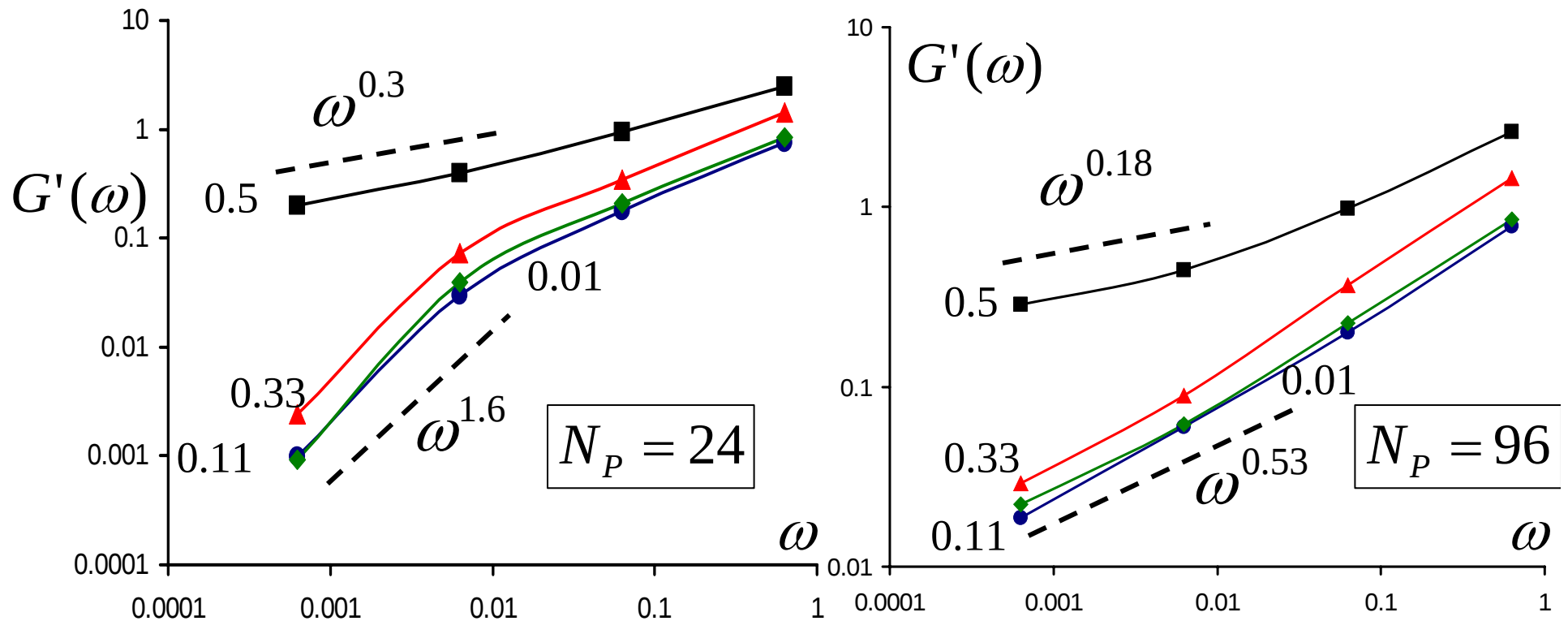
F_{PP}



FENE Springs

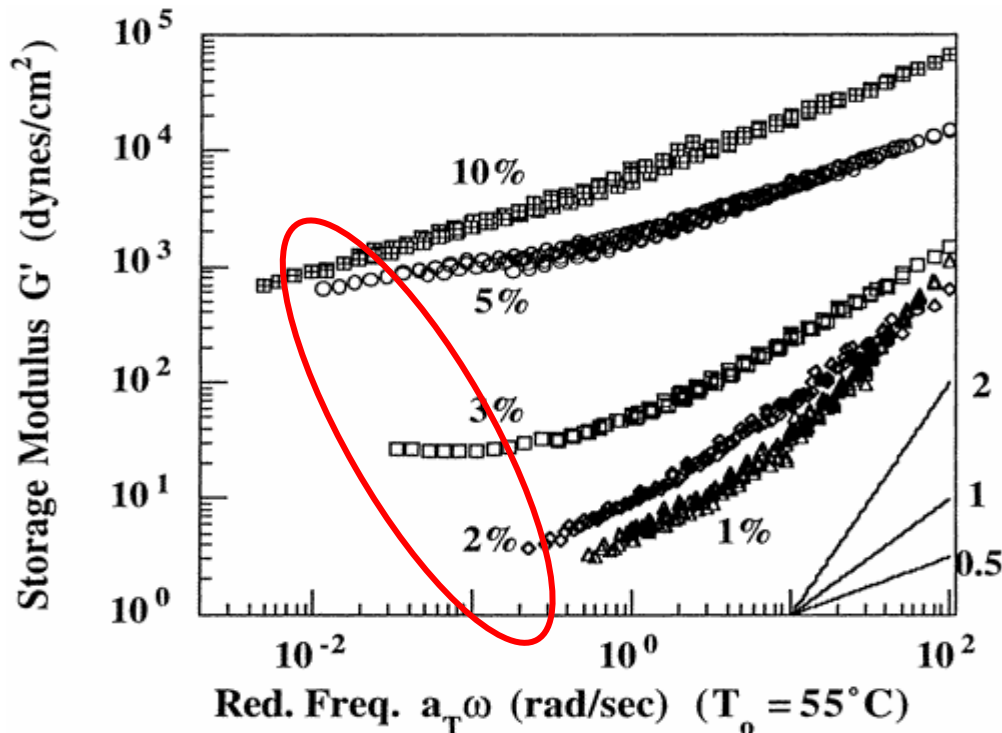
(VP & VG: Macromolecules, 2006; J. of Rheology)

Rheology of PNCs: Linear Viscoelasticity

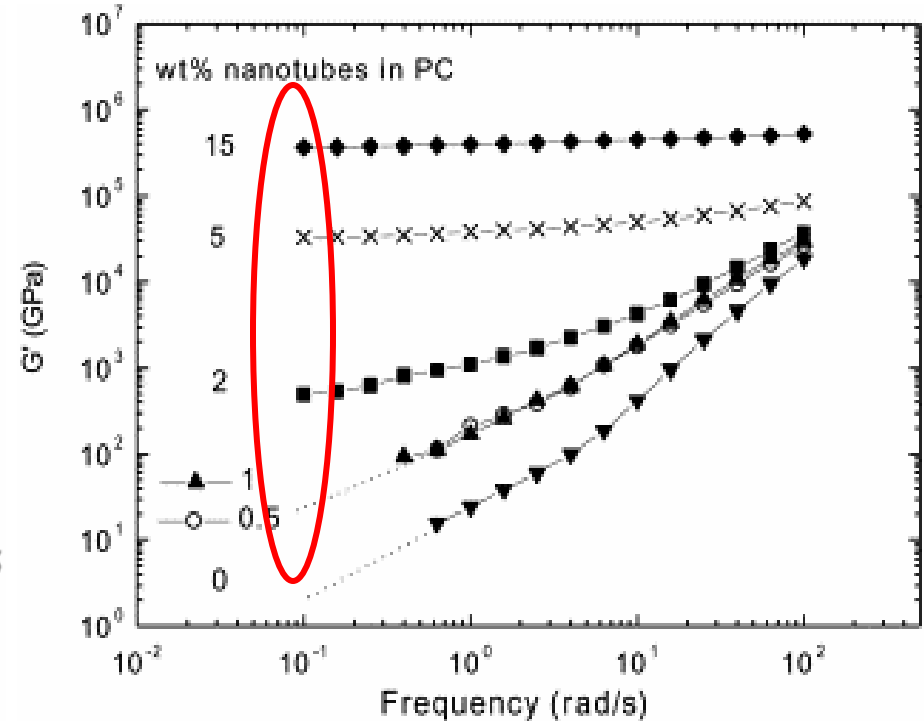


- **Lower Loadings:** Enhancement in modulus but no apparent change in relaxation behavior.
- **Higher Loadings:** Significant enhancement in modulus and a solid-like behavior.

Rheology of PNCs: Linear Viscoelasticity



Polycaprolactone + Montmorillonite Clays*

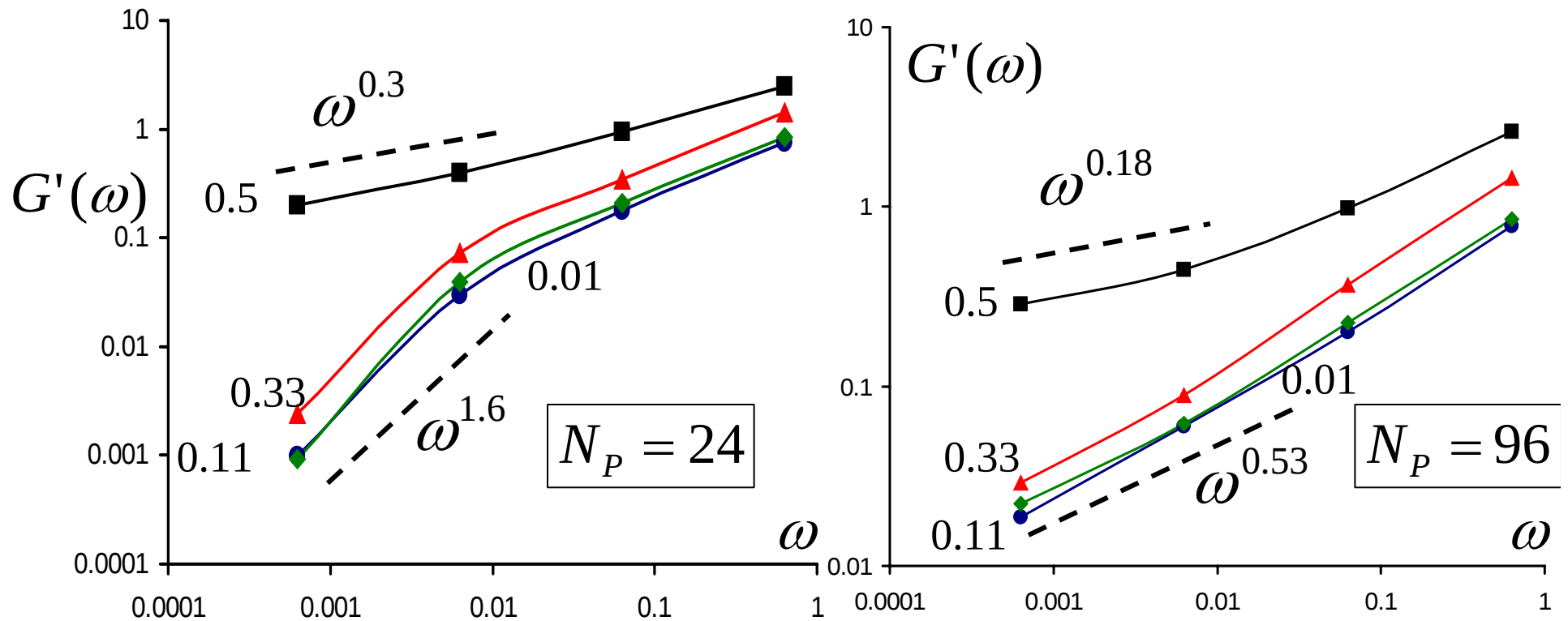


PC + Nanotubes**

- Significant enhancements in elasticity at extremely low loadings
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*:Krishnamoorti and Giannelis; **: Fornes and Paul

Rheology of PNCs: Linear Viscoelasticity

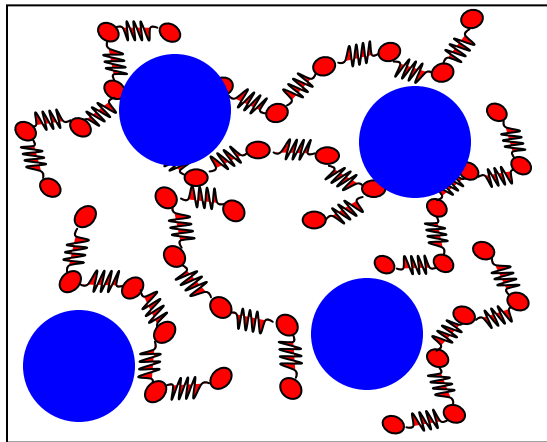


- Enhancement in modulus but no apparent change in relaxation behavior at lower loadings: **Why ?**
- Significant enhancement in modulus and a solid-like behavior at higher loadings: **Why ?** (Not in this talk)

Impact on Polymer Dynamics

- Significant impact upon glass transition temperature and polymer dynamics on adding nanoparticles.#

How does the polymer dynamics change due to addition of nanoparticles ?



$$R(t) \xrightarrow{\text{Normal modes}} X_m(t)$$

$$\downarrow$$

$$\langle X_m(t) X_m(0) \rangle$$

For unentangled polymers,

$$\langle X_m(t) X_m(0) \rangle \approx \exp(-t / \tau_m)$$

$$\tau_m \propto m^{-2}$$

$$\tau_1 N_p^{-2} \propto \xi$$

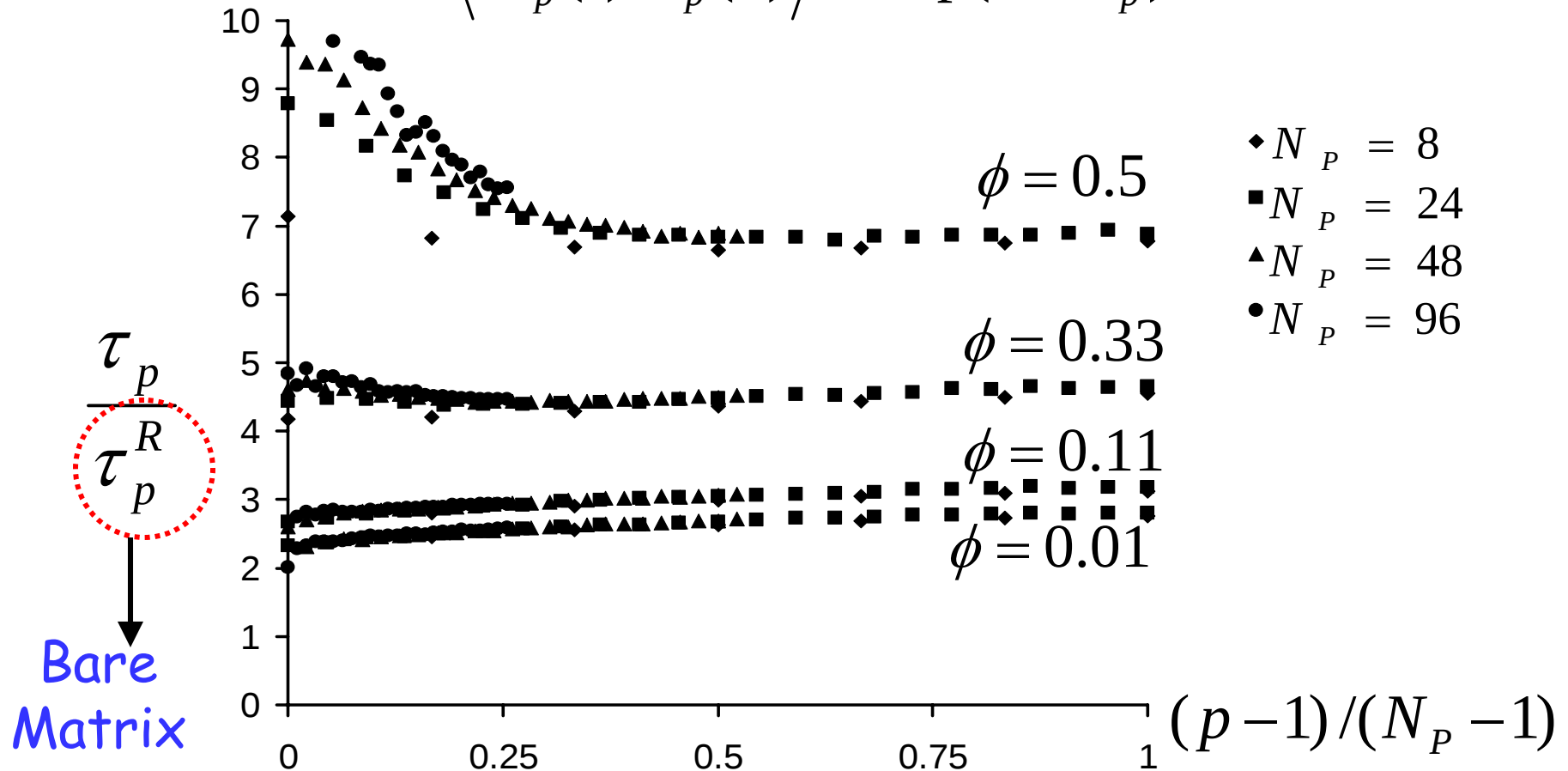
Related to viscosity of media

Simulation Features
 No glass transition
 Coarse-grained model

#: Giannelis, Adv. Pol. Sci, 138, 107

Effect of Particles on Polymer Dynamics

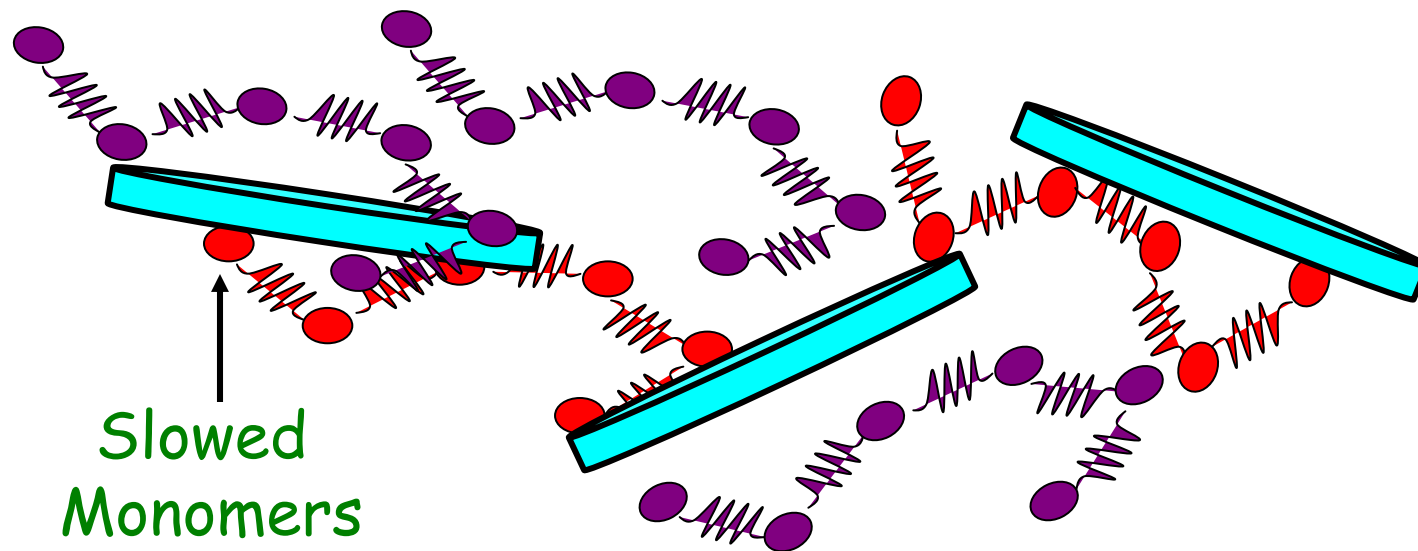
$$\langle X_p(t) X_p(0) \rangle \approx \exp(-t / \tau_p)$$



Overall Slowing of Polymer Relaxations

Effect of Particles on Polymer Dynamics

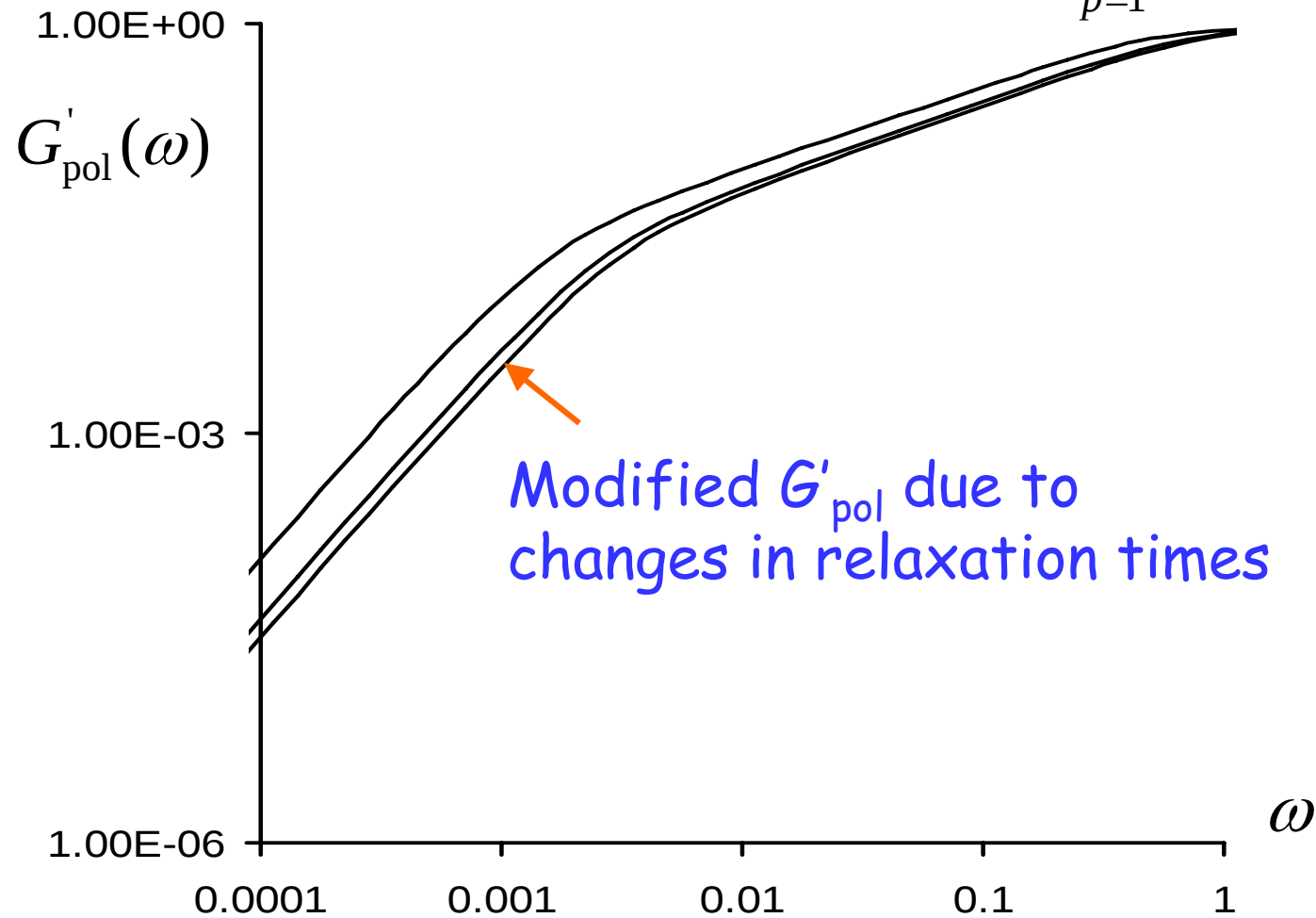
$$\langle X_p(t) X_p(0) \rangle \approx \exp(-t / \tau_p)$$



- Simulations of Grant Smith: Attractions lead to only a weak slowing down in melts.
- Different monomers of a chain access the slower regions:
Overall slowing down of the polymers

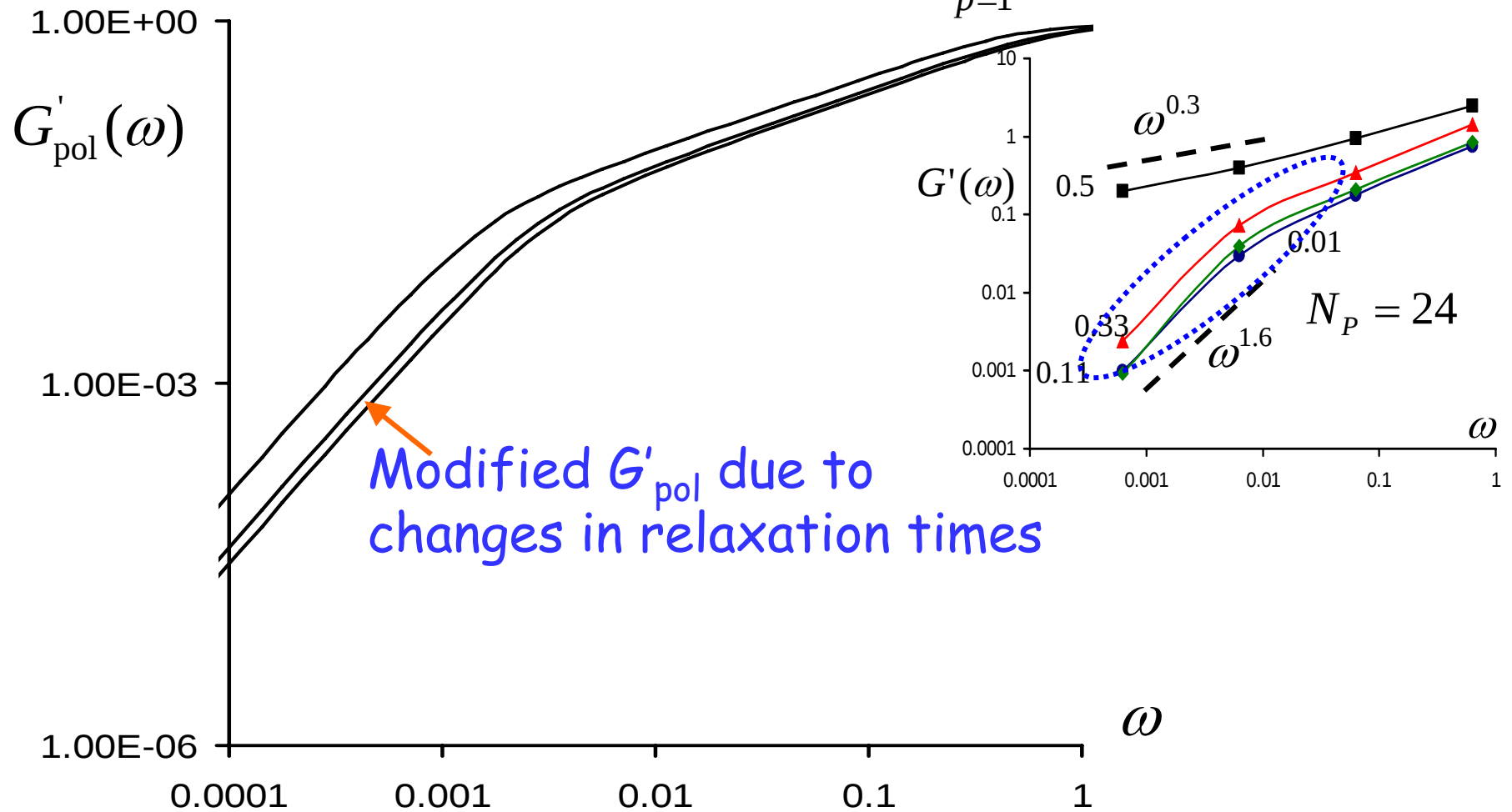
Effect of Particles on Polymer Viscoelasticity

$$\langle X_p(t) X_p(0) \rangle \approx \exp(-t / \tau_p) \Rightarrow G(t) \propto \sum_{p=1}^{N_p} \exp(-2t / \tau_p)$$



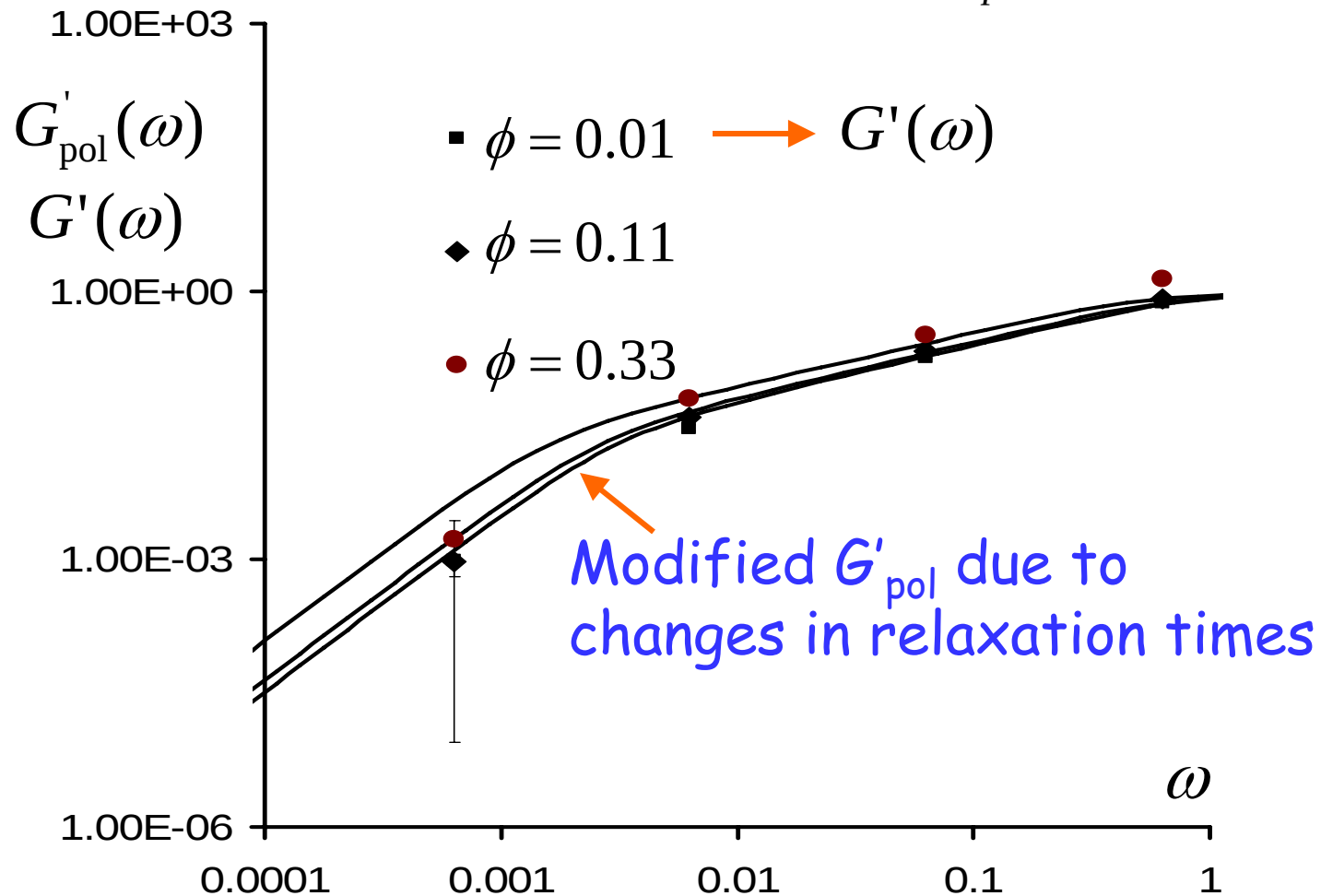
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Effect of Particles on Polymer Viscoelasticity

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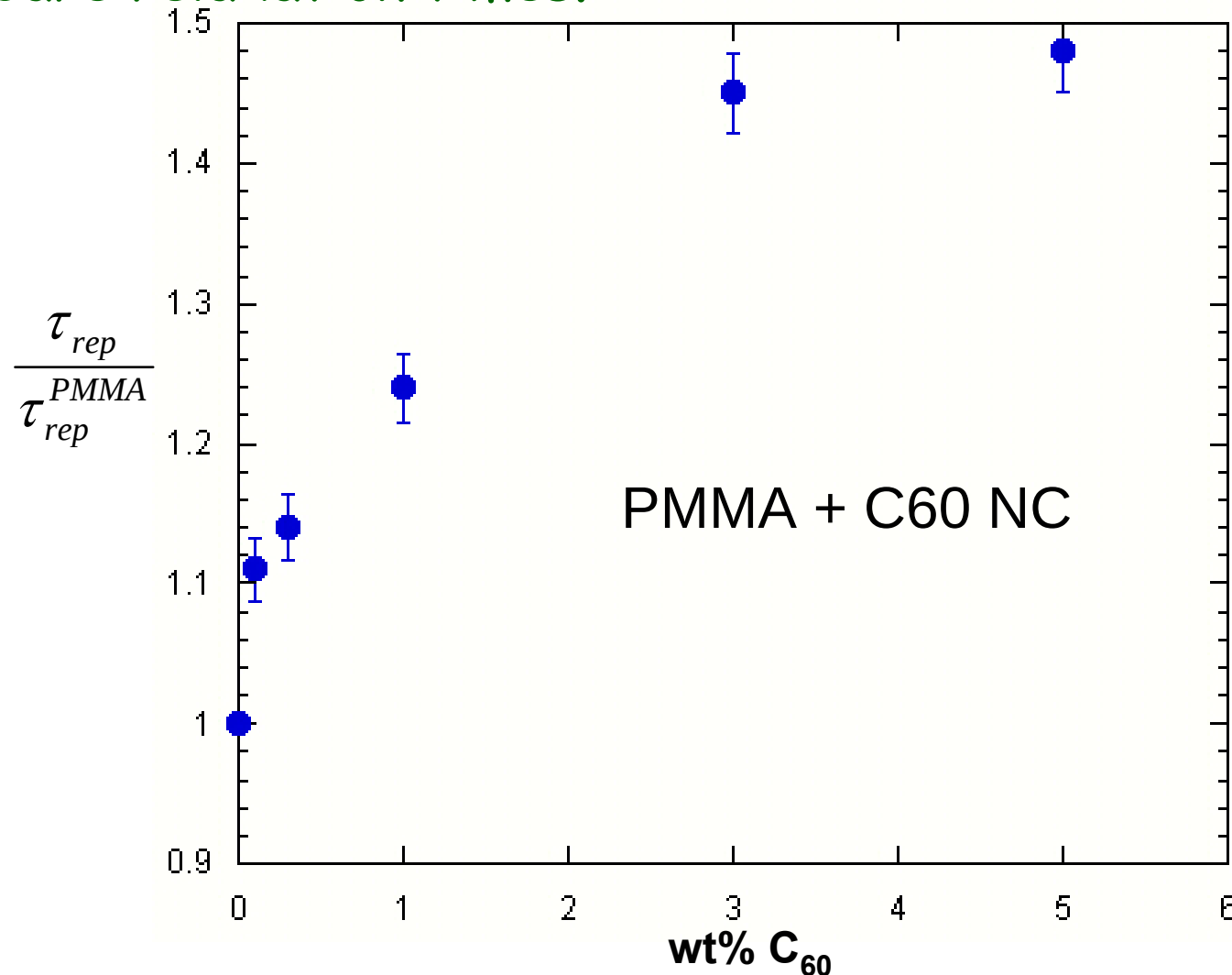


Physical Picture of Polymer Rheology at Low Loadings

- Particle-induced changes in polymer dynamics is responsible.
- For the weakly attractive particles, the above manifests as just a change in relaxation times.
- For strongly attractive particles, the above manifests as the modulus due to polymer-bridged networks.

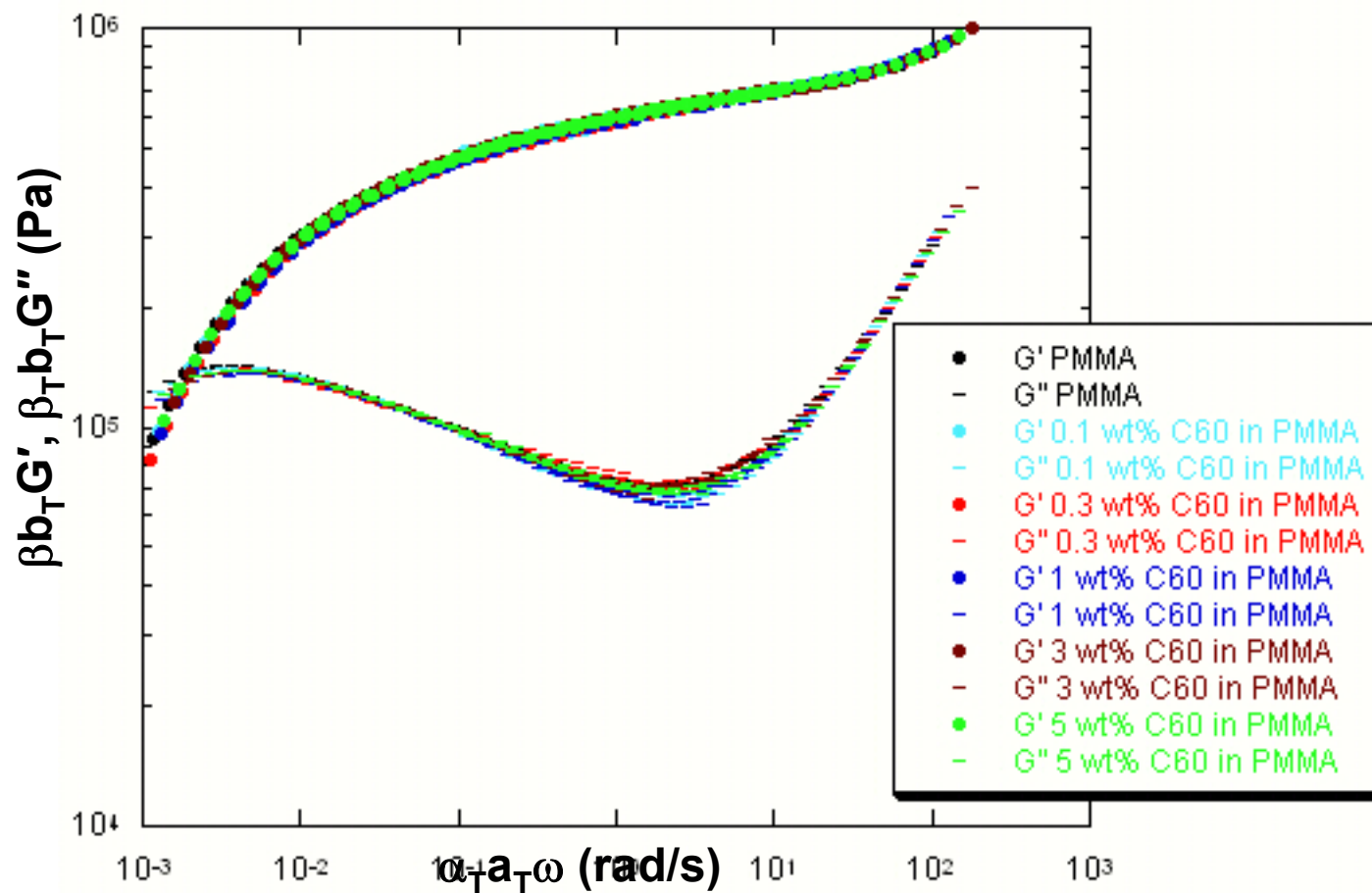
Comparisons to Experiments

- (Kropka, Green and Ganesan, Macromolecules, In Press)
- Comparison of the relaxation times in nanocomposites to the bare relaxation times.



Comparisons to Experiments

- (Kropka, Green and Ganesan, *Macromolecules*, in Press)
- Superposition of mechanical moduli after renormalization of relaxation times.



Rheology of PNCs: Issues and Questions

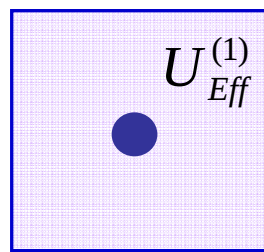
- How do nanoparticles modify the rheology of polymer matrices ? 
- What are the mechanisms underlying the above effects ? 
- At low particle loadings, polymer-bridging of particles is the responsible mechanism.
- What are the parameters governing the mechanisms ?
- What are the elastic and structural properties of the gels ?
- Why do nanoparticles lead to prevelant gelation ?
- How does the concentration of particles and polymer affect the gelation, stability characteristics of the mixture ?

Outline of Approach

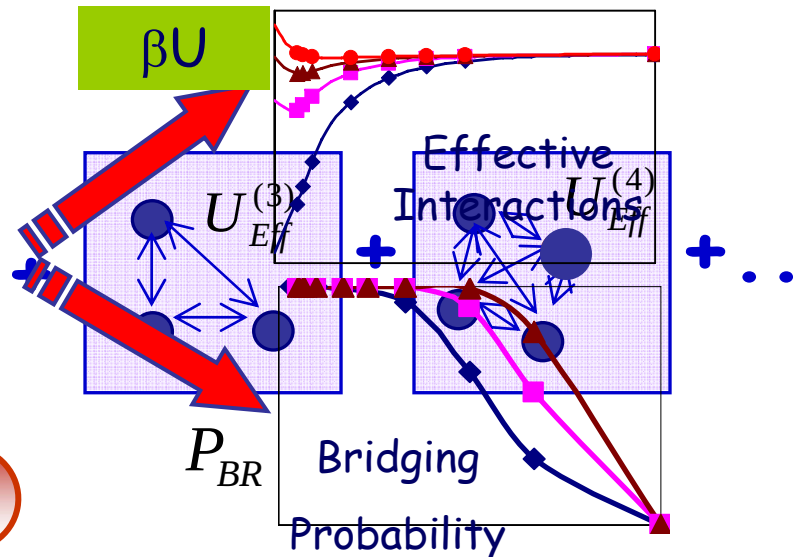
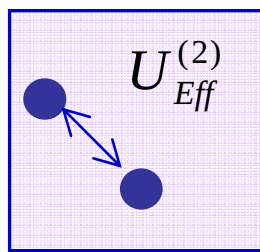
Integrate out polymer degrees of freedom



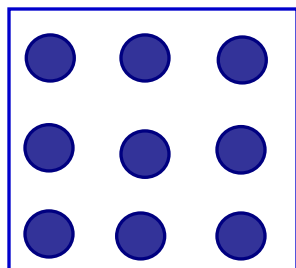
$\mathbb{Z}$



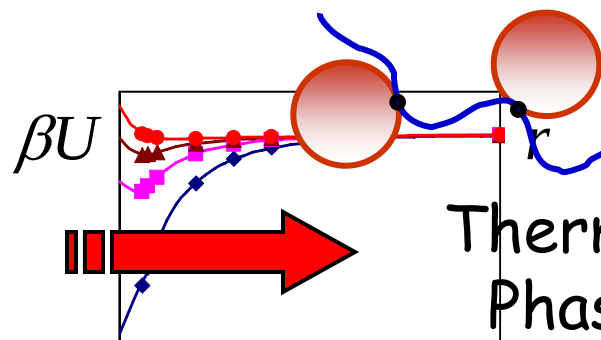
+



Z_P

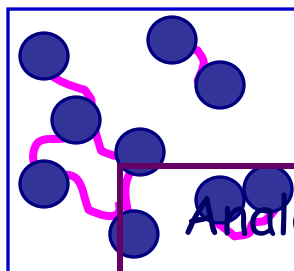


Monte Carlo
And Equilibration



Thermodynamics,
Phase Behavior

MS, VP, VG:
JCP 2005; Langmuir '06;
Macromolecules '06.



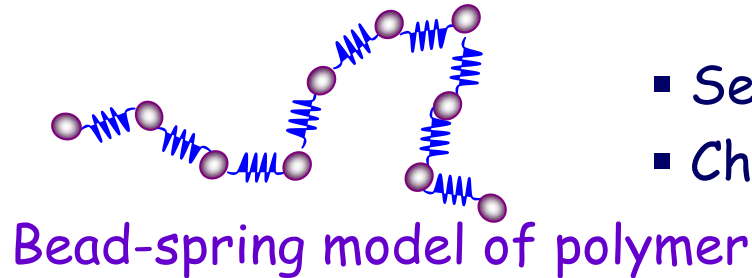
Cluster Counting

Gelation,
Elastic Modulus

Analogous to density functional theories used to obtain
interatomic potentials in quantum chemistry

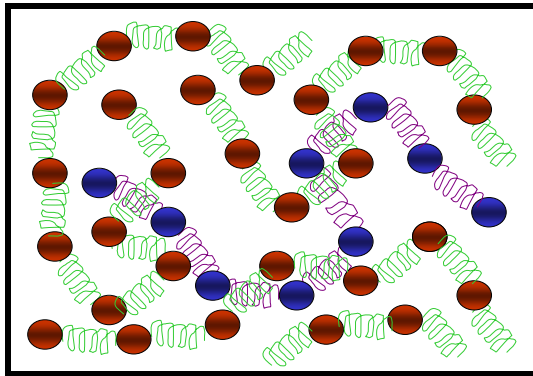
Integrating Out the Polymer By Mean-Field Theory

N segment chain



- Self avoiding random walk
- Charge effects

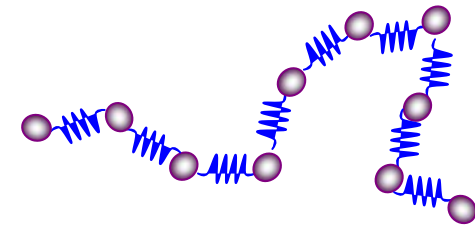
Idea behind mean-field theory*



Thermodynamics



$W(r)$

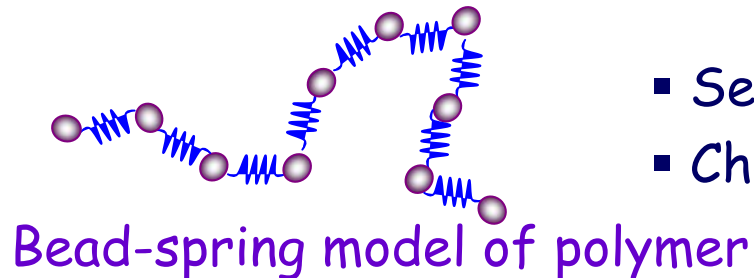


- Single chain in a potential field $W(r)$.
- $W(r)$ determined self-consistently to match statistical properties of polymers, say, the volume fraction.

*: Helfand (1975)

Integrating Out the Polymer By Mean-Field Theory

N segment chain



- Self avoiding random walk
- Charge effects

Mean Field Approach

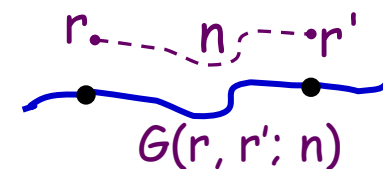
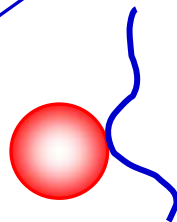
Presence of other chains
Self consistent potential field $w(r)$

Configurations of a chain subjected to $w(r)$

$$\frac{\partial G(r, r'; n)}{\partial s} = \nabla^2 G(r, r'; n) - i w(r) G(r, r'; n) \quad G(r, r'; 0) = \delta(r - r')$$

BCs :

Polymer - particle interactions



Field-Theory Model For Polymers

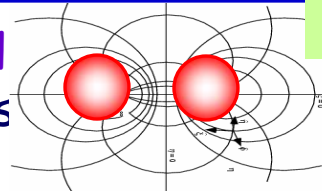
Density distributions
for polymer

Polymer mediated
effective interactions

Numerical solution

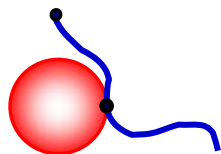
Curvature of particles
accounted exactly!!!

Bispherical
coordinates

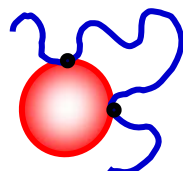


Configurations of a chain subjected to $w(r)$

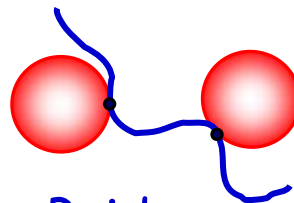
$$\frac{\partial G(r, r'; n)}{\partial s} = \nabla^2 G(r, r'; n) - iw(r)G(r, r'; n) \quad G(r, r'; 0) = \delta(r - r')$$



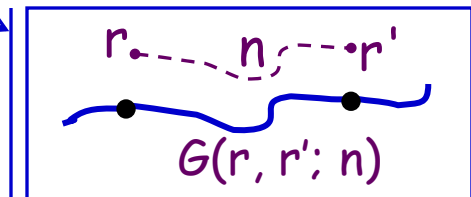
Tail



Loop

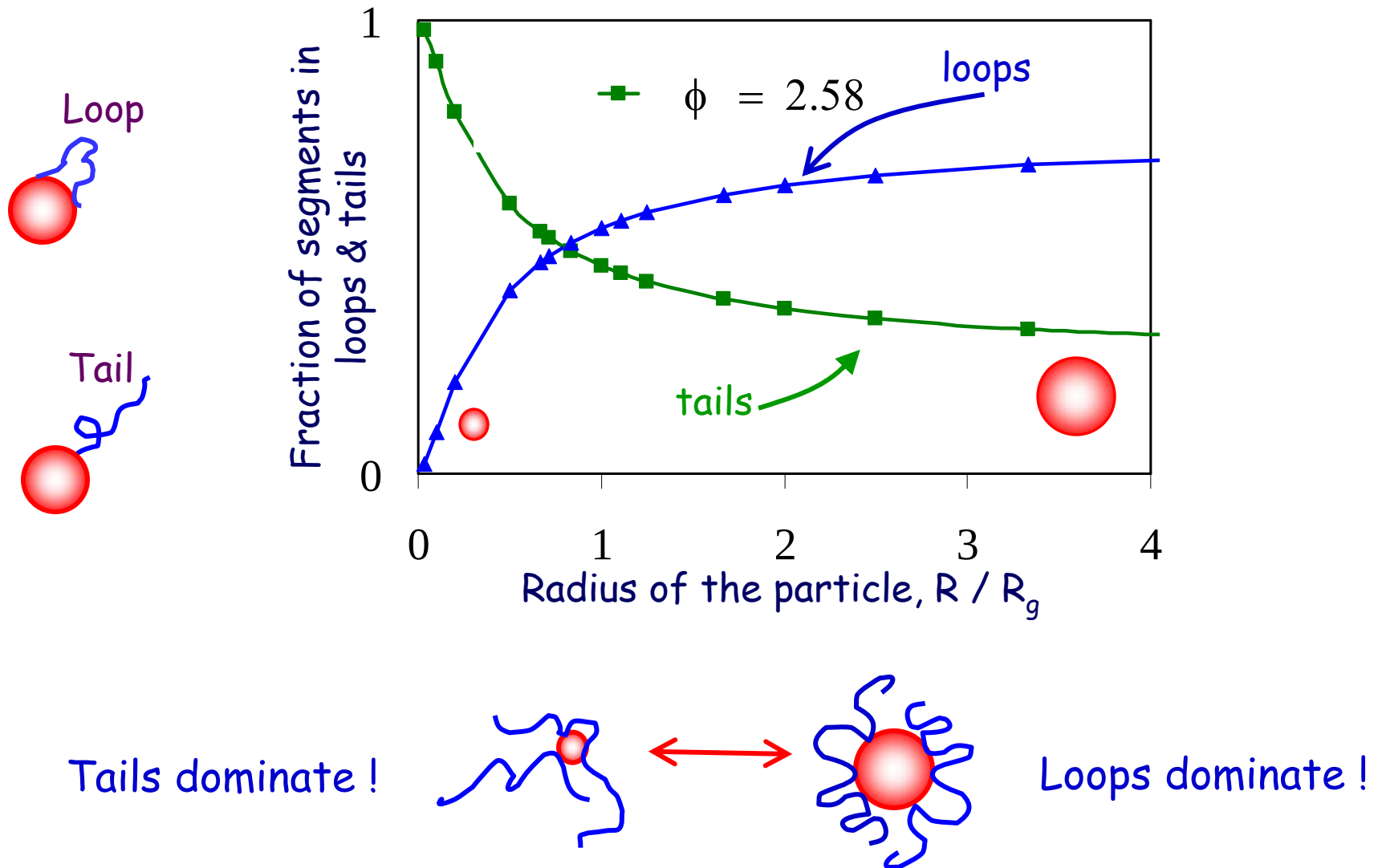


Bridge



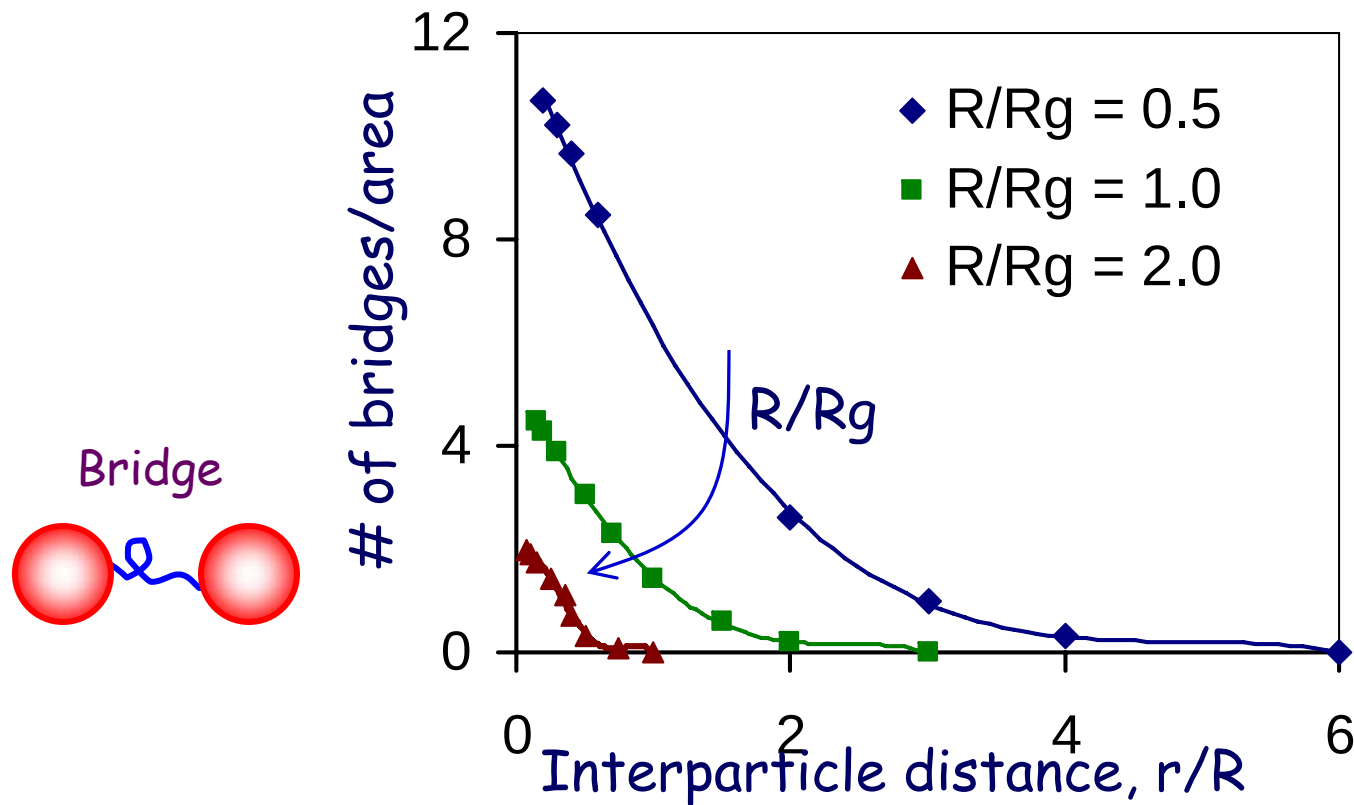
Number and probability distribution

Adsorbed Layer: Particle Size Effects



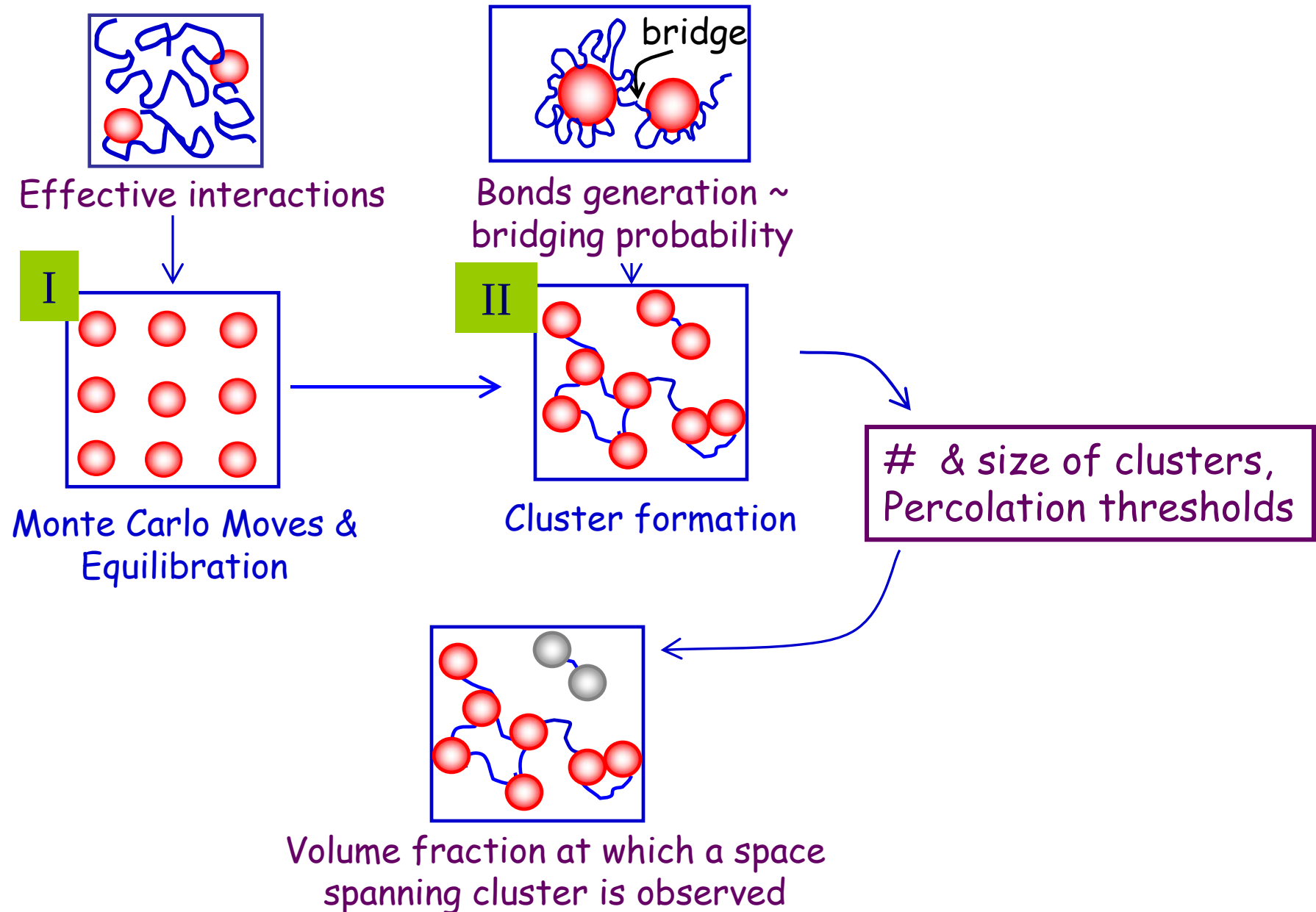
Smaller particles - tails dominate

Bridging: Particle Size Effects



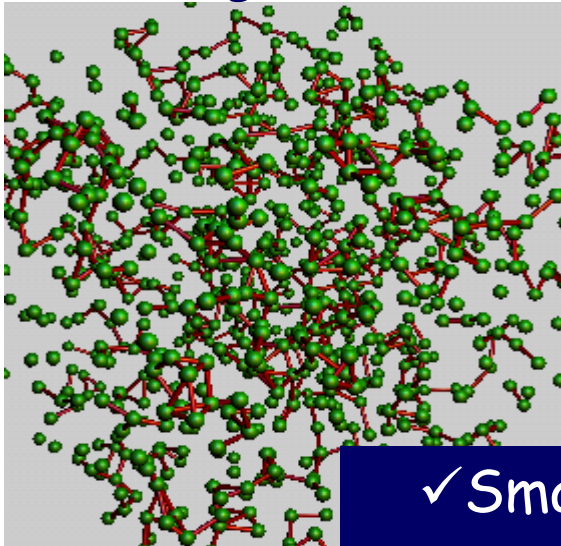
Smaller particles - More number of bridges

Cluster Statistics and Gelation

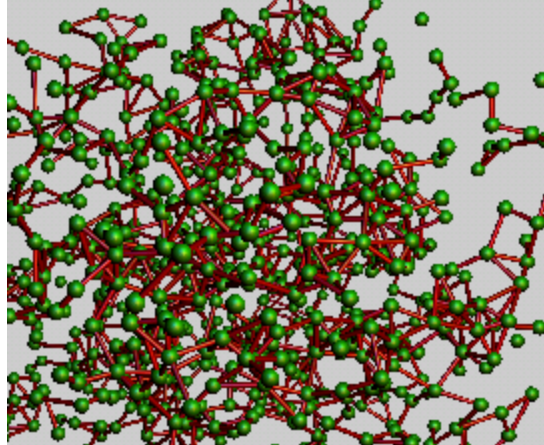


Cluster Statistics and Gelation

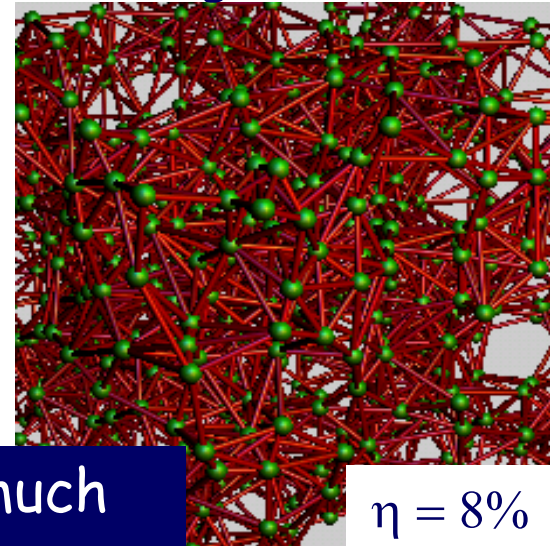
$R/R_g = 2.0$



$R/R_g = 1.0$

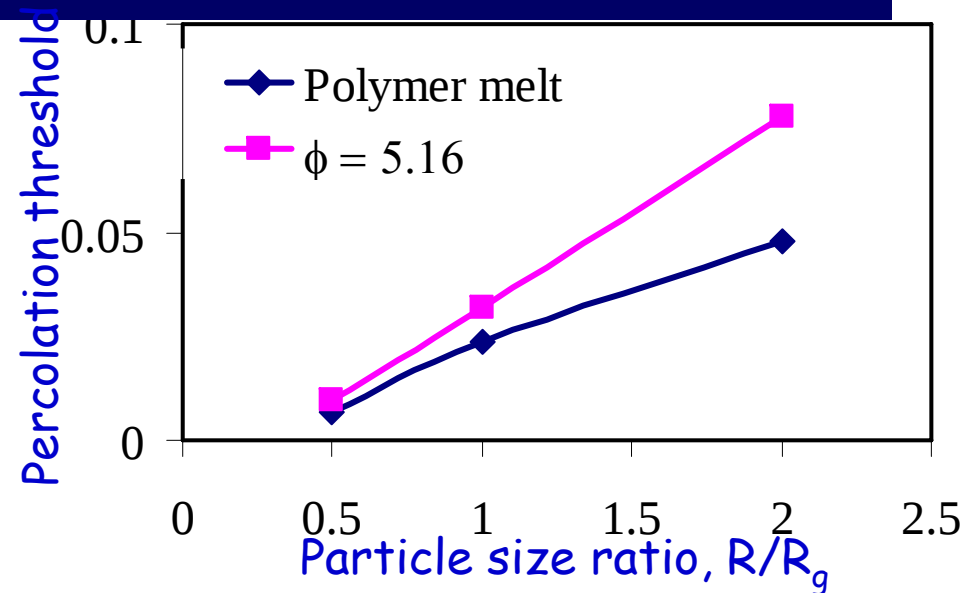


$R/R_g = 0.5$



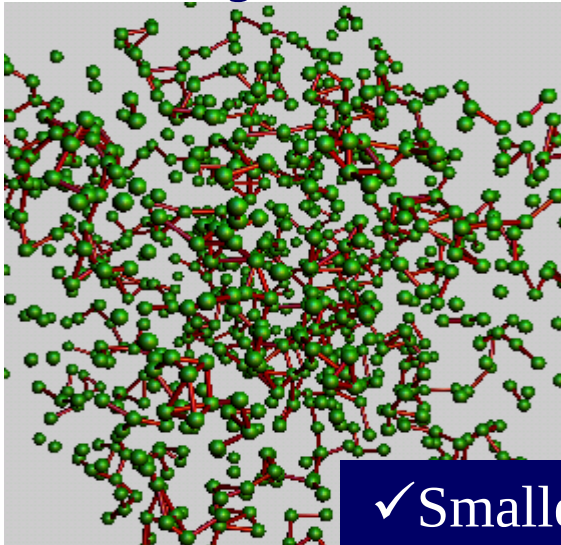
$\eta = 8\%$

✓ Smaller particles - Gel much earlier!

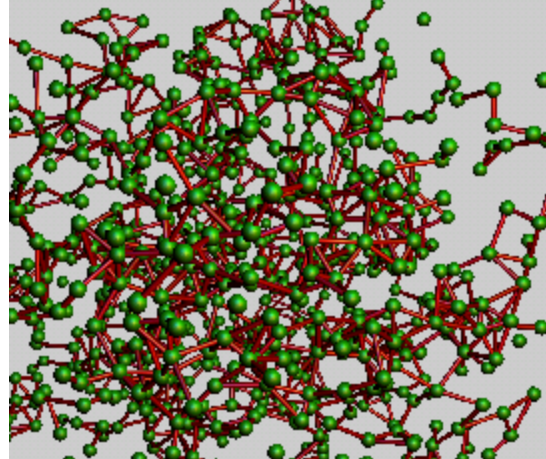


Determining Elastic Properties

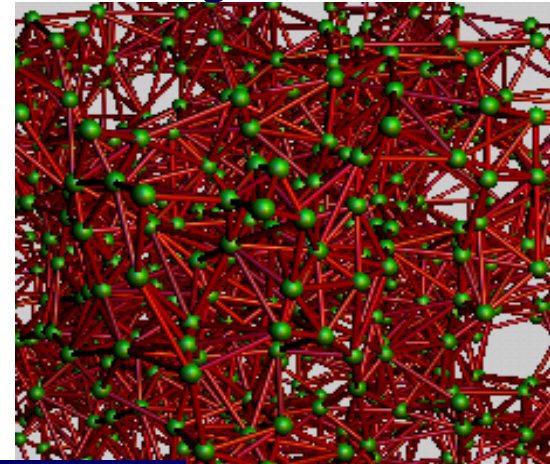
$R/R_g = 2.0$



$R/R_g = 1.0$



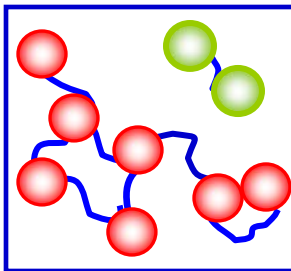
$R/R_g = 0.5$



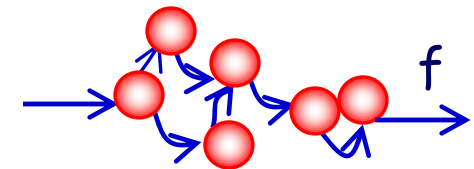
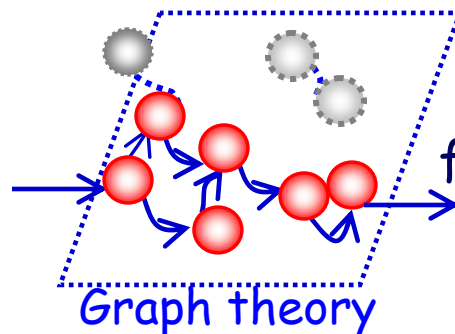
✓ Smaller particles – Gel much earlier!

$\eta = 8\%$

Post-gel systems



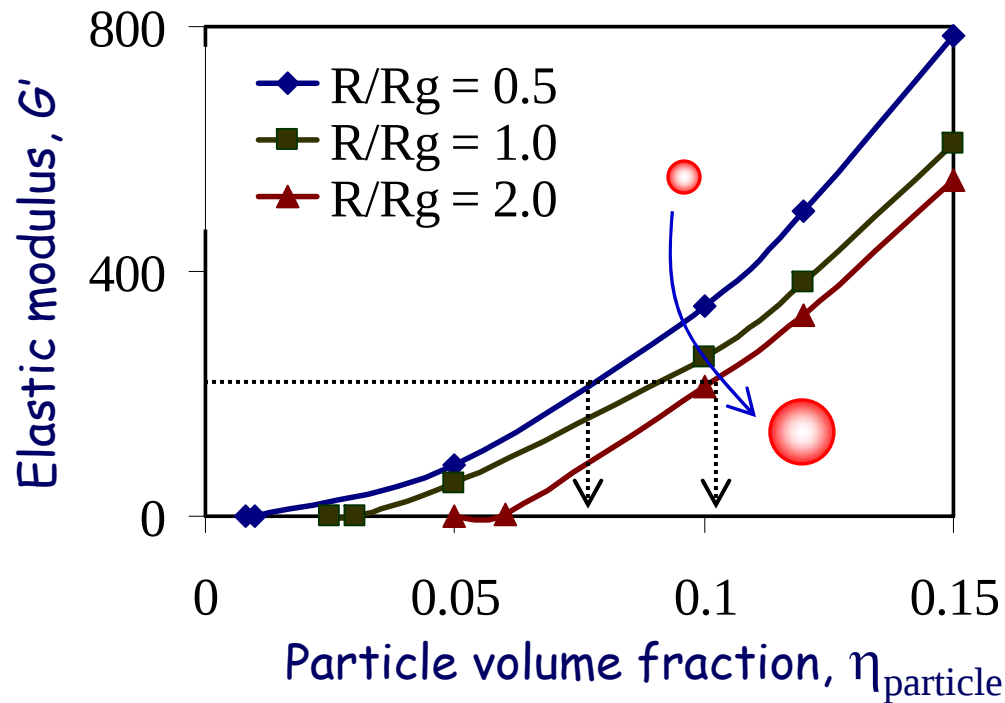
Identification of backbone



Elastic properties

Simple Network Theories: Elastic Moduli = Number of Bridges in backbone

Elastic Moduli of Gels



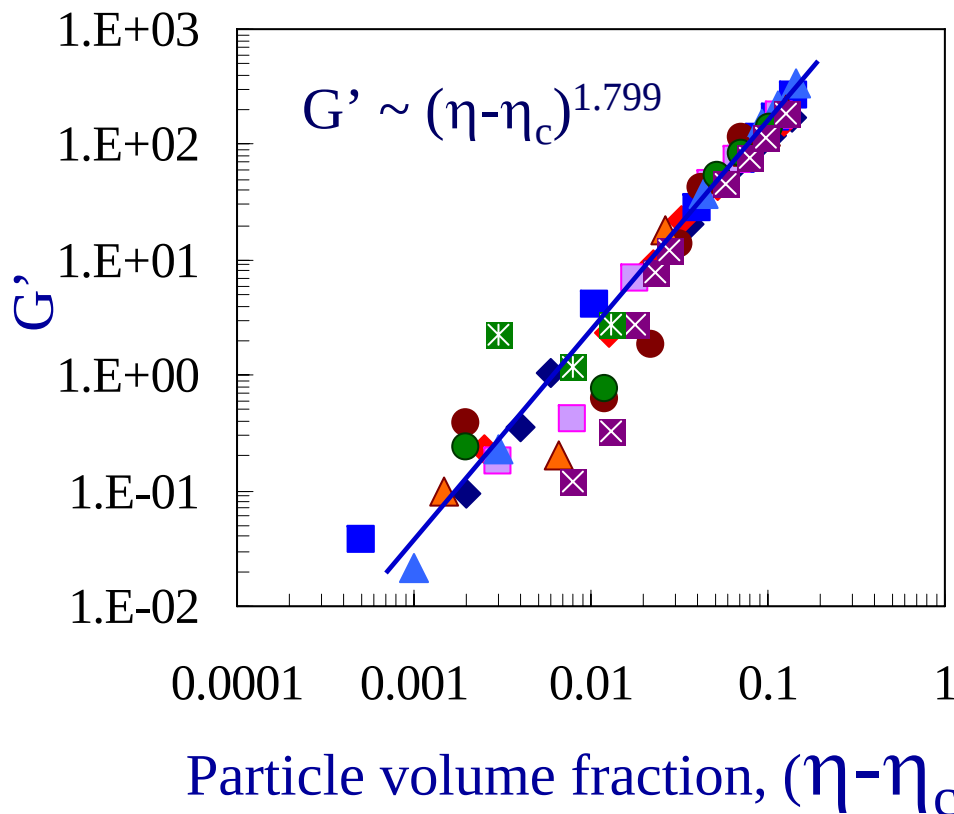
- ✓ Smaller particles $\rightarrow$ stronger reinforcement.
- ✓ Smaller quantities of nanoparticles are required.
- ✓ Bridging induced clustering of particles responsible for reinforcement.

*Surve, Prymitysyn, Ganesan, Phys. Rev. Lett.

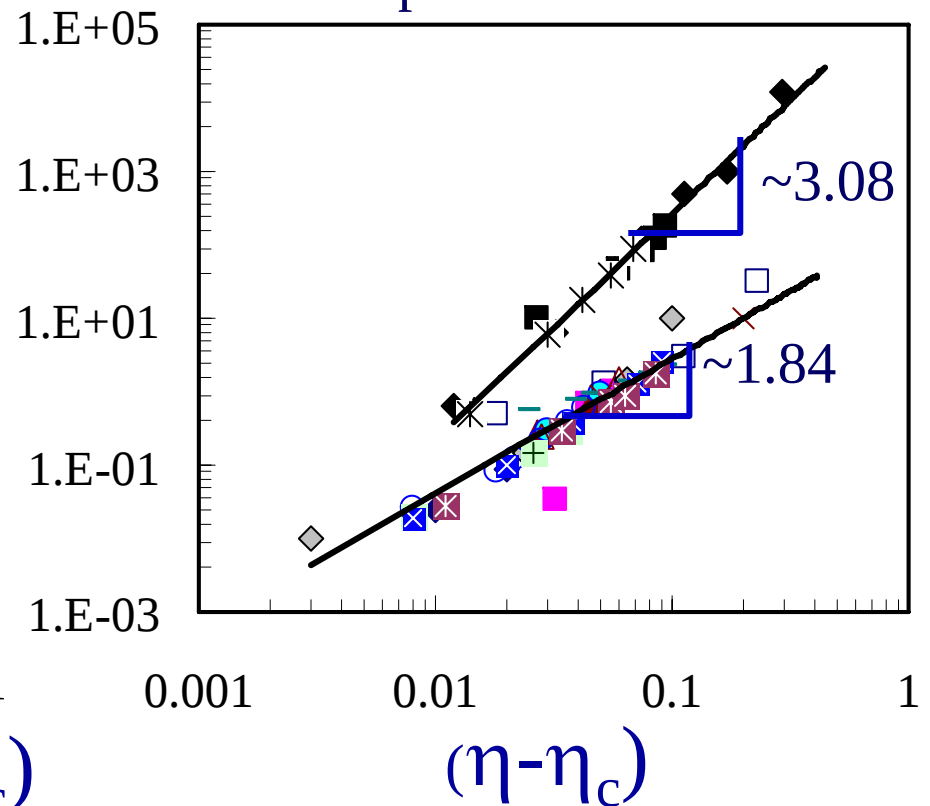
Much Stronger Enhancements of Moduli in Smaller Particles

Scaling of Elastic Moduli of Gels

Simulation results*



Experimental data



*Surve, Prymitysyn, Ganesan, Phys. Rev. Lett.

Universal scaling of elastic modulus

Conclusions - Part II

✓ Gelation

- ✓ For smaller particles, gelation occurs at very low volume fractions
- ✓ Small particles → dense networks
- ✓ Small particles → Much stronger enhancements in moduli

